Elimination of Potential-Induced Degradation for Crystalline Silicon Solar Cells

by

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ABSTRACT

Potential-Induced Degradation (PID) is an extremely serious photovoltaic (PV) durability issue significantly observed in crystalline silicon PV modules due to its rapid power degradation, particularly when compared to other PV degradation modes. The focus of this dissertation is to understand PID mechanisms and to develop PID-free cells and modules.

PID-affected modules have been claimed to be fully recovered by high temperature and reverse potential treatments. However, the results obtained in this work indicate that the near-full recovery of efficiency can be achieved only at high irradiance conditions, but the full recovery of efficiency at low irradiance levels, of shunt resistance, and of quantum efficiency (QE) at short wavelengths could not be achieved. The QE loss observed at short wavelengths was modeled by changing the front surface recombination velocity. The QE scaling error due to a measurement on a PID shunted cell was addressed by developing a very low input impedance accessory applicable to an existing QE system.

The impacts of silicon nitride (SiN_x) anti-reflection coating (ARC) refractive index (RI) and emitter sheet resistance on PID are presented. Low RI ARC cells (1.87) were observed to be PID-susceptible whereas high RI ARC cells (2.05) were determined to be PID-resistant using a method employing high dose corona charging followed by time-resolved measurement of surface voltage. It has been demonstrated that the PID could be prevented by deploying an emitter having a low sheet resistance (~ 60 Ω /sq) even if a PID-susceptible ARC is used in a cell. Secondary ion mass spectroscopy (SIMS) results suggest that a high phosphorous emitter layer hinders sodium transport, which is responsible for the PID. Cells can be screened for PID susceptibility by illuminated lock-



in thermography (ILIT) during the cell fabrication process, and the sample structure for this can advantageously be simplified as long as the sample has the SiN_x ARC and an aluminum back surface field.

Finally, this dissertation presents a prospective method for eliminating or minimizing the PID issue either in the factory during manufacturing or in the field after system installation. The method uses commercially available, thin, and flexible Corning® Willow® Glass sheets or strips on the PV module glass superstrates, disrupting the current leakage path from the cells to the grounded frame.



To my God



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Chapter 1

INTRODUCTION

1.1 Reliability of Photovoltaics

A photovoltaic (PV) cell is a semiconductor device that converts sunlight into electricity. Since the PV effect was discovered by Becquerel in 1839 [1], a variety of PV cell technologies have been developed, for example, single junction crystalline silicon, thin film, such as cadmium telluride (CdTe) or copper indium gallium deselenide (CIGS), and multi junction concentrator PV (CPV). Currently in 2016, crystalline silicon dominates the solar market with more than a share of 90% as shown in Figure 1.1 [2].

Since the PV cell needs sunlight to generate electricity it is necessary to be operated outdoors. However, PV cells operated outdoors would be directly exposed to severe environmental condition, such as high temperature and humidity. It has been well known that the temperature and humidity are some of the important factors determining performance and reliability of semiconductor devices. This applies to PV cells as well. Therefore, PV cells are generally encapsulated with additional materials, such as glass and backsheet, to protect the PV cells from such environments, and these encapsulated constructions are called PV modules. For typical crystalline silicon PV modules, an aluminum frame is usually attached around the edge of a series of connected cells for the purpose of mounting and protection. Multiple PV modules are installed on the roof of a house/building or in a field for a maximum sunlight absorption. Especially, a large number of PV modules connected in series or parallel and operated in a field is referred to as a PV system. The aluminum frames of those PV modules in a PV system should be grounded for safety reasons. The grounding in a PV system that is up to 1500 Vdc causes



very high voltage stress to the cells in PV modules. In addition to the high voltage in a PV system, temperature, humidity, and ultraviolet (UV) intensity are the important factors governing the reliability of PV cells or modules. Whereas the aforementioned three factors are dependent on the season and the region where the PV modules are installed and operated, the high voltage factor differs from country to country due to varying electrical regulations. Figure 1.2 - 1.4 show various environments around world. It would be expected that the PV reliability could be a critical issue in regions where extreme temperature, humidity, and UV intensity are present, for example, countries in South America.



Figure 1.1 Percentage of Global Annual Production [2]





Atlas of the Biosphere Center for Sustainability and the Global Environment University of Wisconsin - Madison

Figure 1.2 A World Map of Average Annual Temperature [3]



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Figure 1.3 A World Map of Average Annual Relative Humidity [4]





Figure 1.4 A World Map of Solar UV Index [5]

Reliability is one of the key factors in reducing the cost of the PV system. In many instances, a PV system still costs more than traditional energy sources [6-8]. The PV energy would be more cost effective by increasing the reliability, thereby leading to longer lifetime. Therefore, PV reliability is getting more attention from researchers, industries, and policy makers.

Owing principally to its large market share, many reliability issues have been observed on crystalline silicon PV modules. Those reliability issues are not only from PV cells but also relate to PV module materials, such as glass, encapsulation material (e.g., EVA: Ethylene Vinyl Acetate), backsheet and ribbons. PV module material quality is directly related to the ultimate reliability of PV cells/modules since these materials are closely contacted with the cells. Therefore, a reliability analysis of PV also needs to be approached from the perspective of PV modules. Figure 1.5 shows typical failure scenarios of crystalline silicon PV modules [9]. They are infant-failure, midlife-failure, and wear-out-failure. It should be noticed that highest degradation occurs in infant-failure.



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This is due to flawed PV cells/modules, and these issues should be fixed immediately once they are observed. Midlife-failure is typically related to encapsulation materials, such as EVA or glass, and this failure period lasts until the manufacturer's warranty year. Wear-out failure occurs at the end of life and is not covered by the manufacturer's warranty. Some common reliability failure modes of crystalline silicon PV modules are presented below.



Figure 1.5 Three Typical Failure Scenarios for Wafer-Based Crystalline PV Modules [9]

a. PV Module Material Degradation

As it was mentioned previously, a PV module consists of several packaging materials, such as glass, encapsulant, and backsheet. The typical superstrate that has been used for PV modules is a glass that is tempered, low-iron content, and 3.2 mm thickness. One obvious glass degradation mechanism is breakage which, in turn, might cause output power loss and safety problems [10]. Usually, high impacts, such as from rocks, hail or a



bullet, could break the glass. Also, very high loads, such as caused by snow on the module surface and strong wind, can lead to glass breakage.

It has been known that polymer encapsulants which are used to laminate glass to solar cells in a module are degraded by ultraviolet (UV) light and heat. The discoloration of the encapsulation material by UV and heat was usually observed on long-term, outdoor-operated PV modules, as shown in Figure 1.6. Therefore, it is desirable for the module glass to prevent UV sunlight that is lower than 400 nm from reaching the encapsulant material. One solution is to add a small amount of cerium (Ce) to glass formulations in order to filter UV light [11]. Figure 1.7 shows that UV screening is more effective when Ce is added to glass than comparable glass without Ce.



Figure 1.6 Discoloration of the Encapsulant due to UV [12]





Figure 1.7 Transmittance of Module Superstrate Materials Including Glass Samples with and without Ce Additive [11].

b. Adhesional Degradation

Since PV modules are laminated with glass, encapsulant, and backsheet, severe environmental conditions, such as high temperature and humidity, can lead to of PV module delamination. Delamination is the breakdown of the bonds between material layers in a PV module [13]. It has been shown that front side delamination (Figure 1.8), the two forms of which are glass/encapsulant and encapsulant/cell, is more common than back-side delamination i.e., encapsulant/backsheet [13]. Front-side delamination causes output power degradation due to lower transmittance from the glass and leads to poor heat dissipation, which increases reverse-bias cell heating. In addition, delamination increases the probability of moisture ingress which leads to corrosion and an increase in leakage current. The corrosion of cell metallization due to moisture ingress in a PV module causes lower output power from the module as well as increase electrical safety problems.





Figure 1.8 PV Module Showing Front-Side Delamination [12]

c. Interconnect and Solder Bond Degradation

In order to connect cells in series or parallel as a module, the cells are connected with ribbon and soldered. It has been observed that thermal expansion and contraction or heavy repeated stress on the module causes interconnect breaks as shown in Figure 1.9 [14]. It can be prevented by using thinner ribbon or a backsheet that has lower thermal expansion coefficient [14].

The solder bonds can fail because of stresses caused by a repetition of high temperature and high humidity in the field. Also, when modules are stressed by a heavy load such as snow, solder experiences more fatigue which eventually causes a decrease of cell/module output power. Low solder bond strength has been observed from long-term outdoor exposure of crystalline silicon PV modules, and the solder bond strength depends on the solder type and process as well as on the optimization of both the screen printing and solder bond processes [15]. Also, voids in poor bonds cause the high possibility of



water ingress leading to corrosion and low solder bond strength when PV modules are operated outdoors [15].



Figure 1.9 Broken Interconnect [14]

1.2 Motivation

Several common reliability issues observed on crystalline silicon PV cells/modules were presented in the previous section. Some of them have been significantly reduced, as there are many solutions for those issues that have been researched for a long time. In 2005, a new reliability issue, which is called Potential-Induced Degradation (PID), was introduced and got attention from many PV people because of its rapid degradation, which is in the form of a huge output power drop (>40%) in a very short period of time (2-3 months). The PID was mainly observed on crystalline silicon PV modules in a PV system. This could be a huge problem. As it was previously shown in Figure 1.1, crystalline silicon has been a major technology (>90%) for a couple of decades. Moreover, 91% of PV modules around the world have been installed just between 2008 and 2014 as shown in Figure 1.10. Based on these facts, we cannot imagine how many crystalline silicon PV modules in the world are suffering or in danger



from PID. The PID issue should be addressed for all of PID-experienced PV modules, PID-susceptible PV modules already installed, and future PV cells/modules.



Figure 1.10 Global Solar PV Cumulative Installed Capacity 2000-2014 [16]

Due to its serious impact on PV modules, several methods for avoiding PID have been developed and deployed. However, most of the techniques addressing PID focus on making new PID-free cells/modules. As was mentioned before, numerous PIDsusceptible modules that have not yet experienced PID are operating in a field. A method for preventing PID for these modules should be researched and developed to avoid additional PID degradation.

Although several methods for manufacturing PID-free cells/modules have been developed, a precise physical mechanism of PID remains still unclear. There is a difficulty in analyzing PID-degraded solar cells since it is extremely complicated to take the cell out from laminated modules without breakage. Therefore, a new PID testing



method enabling clean PID sample collection is needed. The new testing method should allow further semiconductor characterization techniques, such as secondary ion mass spectroscopy (SIMS) or scanning electron microscopy (SEM), to be performed for the purpose of further defining the PID mechanism.

1.3 Outline

In Chapter 2, a background of high voltage stress and PID is covered. It presents why PID occurs in a PV system and the status of current research and development. Since the PID mechanism is still unclear, it is necessary to carry out further experiments in addition to the ones done by other researchers to determine the PID mechanism. Chapters 3 through 6 attempt to elucidate PID characteristics as this knowledge is very helpful for eliminating PID degradation from PV modules. In Chapter 3, PID characteristics focused on PID recovery is presented. This chapter covers Quantum Efficiency (QE) loss due to PID and challenges regarding QE measurements on PID-stressed cells (e.g., low shunt resistance cells). The effect of different type of PV module glass on PID is also presented in this chapter. In Chapter 4, a detailed QE loss analysis is presented. A new PID testing method enabling QE measurements on bare cells is introduced. The QE loss is modeled by changing the front surface recombination velocity. A PID surface inversion model, one of the proposed PID mechanisms, is verified by the Sentaurus semiconductor simulation program. In Chapter 5, the effect of refractive index of silicon nitride (SiN_x) anti-reflection coating (ARC) on PID-susceptibility is presented as along with an emitter sheet resistance effect. These are two of the key factors in preventing PID at the cell level. PID-susceptibility of the SiN_x ARC is analyzed by using semiconductor characterization techniques. An emitter is characterized by SIMS and electrochemical capacitance voltage



(ECV) in terms of PID. In Chapter 6, a novel technique screening PID-susceptible cells during the solar cell fabrication process is introduced. Illuminated Lock-In Thermography (ILIT) is used for the screening technique. Finally, a PID prevention method that could be applied both during PV module manufacturing and on PV modules already installed and operated in a field is presented in Chapter 7. A Corning Willow Glass strip that does not interfere with the PV module glass transmittance is used. The conclusion and summary of this dissertation are in Chapter 8.



Chapter 2

HIGH VOLTAGE STRESS AND POTENTIAL-INDUCED DEGRADATION

Due to the low-voltage (~ 0.6 V) output power of crystalline silicon solar cells, the cells are connected in a series as a photovoltaic (PV) module. Those cells are laminated in order to protect them from environmental or mechanical stresses. The typical module structure is glass–ethylene vinyl acetate (EVA) (encapsulant)–cell–EVA (encapsulant)–backsheet, as shown in Figure 2.1, and aluminum frames also are attached to the edge of the laminated module, which are used for mounting, grounding, and protection. For example, these days, common monocrystalline silicon modules have 72 cells in a series connection, which has around 200–300 W output power. A few PV modules are sufficient for residential purposes, such as building applied photovoltaics (BAPV). However, when they are used on a utility scale, such as 10s of MW PV power plants, a large number of PV modules are needed to be connected, as shown in Figure 2.2.

In order to meet this large power requirement, PV modules are connected in a series and parallel. In the United States, for the systems with public access, the maximum system voltage is 600 Vdc in accordance with National Electrical Code (NEC), while European countries allow up to 1000 Vdc. Therefore, in these systems, PV modules face high voltage stress (HVS) since the PV module frame is grounded for safety reasons. Cells in the PV module also are exposed to HVS, and eventually the cells, the modules, and the system may experience performance degradation.




Figure 2.1 Common Crystalline Silicon PV Module Structure.



Figure 2.2 Example of (a) PV Cell (~ 0.6 V), (b) PV Module (~ 40 V), and (c) PV System (~ 600 V).

2.1 Electrochemical Corrosion

When PV modules are mounted in a field application, there is a positive/negative potential difference between the cell and frame with respect to ground, depending on the type of grounding in a system. Thus, some cells might be exposed to up to +600 V or -600 V, in accordance with the NEC, which gives rise to low-level leakage currents between the cell and frame in the long term in the field. The leakage current is composed



of charge carriers that react with the encapsulant material, cell, and frame to produce corrosion products [17].

The electrochemical corrosion was first observed in accelerated long-term biastemperature-humidity tests by Wyle Laboratories, and Jet Propulsion Laboratory (JPL) has conducted several experiments related to the electrochemical corrosion [17, 18]. It has been known that the electrochemical corrosion occurs severely when there is a positive potential difference between the cell and module frame with high temperature and relative humidity (RH) conditions. In the 1980s, JPL carried out the experiments in a damp heat chamber, which has 85 °C and 85% RH conditions, in order to predict electrochemical corrosion breakdown of the crystalline silicon module, and life prediction equations were introduced [17]. The test conditions, such as positive potential to the cells with respect to the grounded frame with a high temperature and high humidity, resulted in a dissolution of cell metallization material into encapsulant material, such as EVA, in crystalline silicon PV modules, as shown in Figure 2.3 [19]. The electrochemical corrosion leads to a decrease in the maximum output power (P_{max}) of the module, which mainly could be caused by an increase of series resistance (R_s) due to the corrosion and dissolution of cell metallization [17]. It has been observed recently that the electrochemical corrosion could occur more in the cracked/microcracked cell area [20]. In addition to the electrochemical corrosion, such high positive potential, with respect to the frame in damp heat conditions, could cause a removal of silicon nitride (SiN_x) antireflection coating (ARC) layers, and this was even worse for high negative potential, as shown in Figure 2.4 [19, 21-23].





Figure 2.3 Electrochemical Corrosion of Solar Cell after 2000 h in Damp Heat with +600 V Applied to the Cell with Respect to the Module Frame [19].



Figure 2.4 Removal of the SiN_x ARC after 1000 h in 85 °C/85% RH with -600 V [19].

The electrochemical corrosion has been observed in thin-film modules as well. For amorphous silicon PV modules, a similar behavior of electrochemical corrosion has been observed as in the crystalline silicon modules; however, the degradation is much higher than in the crystalline silicon modules [24]. The corrosion could be reduced by using higher resistivity encapsulant material; for example, using EVA as an encapsulant



for the PV modules shows lower degradation than using polyvinyl butyral (PVB) [24]. The thin film modules have other HVS degradation in addition to electrochemical corrosion. Transparent conductive oxide (TCO) breakage and delamination was observed in cadmium telluride (CdTe) and a-Si solar modules under –600 V potential, but the modules rarely were degraded for +600 V potential modules [25].

Since the electrochemical corrosion is caused by leakage currents due to the potential difference between the cell and frame, minimizing the module leakage current has been regarded as important work in order to minimize the corrosion. According to JPL's paper, the module leakage currents depend on encapsulant material, temperature, and humidity [26]. There are several encapsulant materials that have been used in the PV module industry, for example EVA, PVB, and ionomer. Due to the higher electrical conductivity of PVB than EVA, PVB experiences a higher leakage current at the same temperature and humidity, as shown in Figure 2.5. Also, it has been observed that the electrical conductivity of the encapsulant material is increased as the temperature and RH increase [18, 26]. Therefore, the electrochemical corrosion could be minimized by using encapsulant material that has high-volume resistivity at a high temperature and RH.





Figure 2.5 Electrical Conductivity of PVB and EVA as Function of Temperature and RH [26].

2.2 Polarization

In 2005, the polarization effect was first reported by U.S. solar manufacturer SunPower, which manufactures high-efficiency n-type solar cells. The polarization effect was observed from the n-type cell modules installed at an outdoor field, and a transformerless inverter was used in the system [27]. Since a transformerless inverter that does not allow grounding of a pole was used, half of the modules in a string had a negative voltage relative to ground, and half of them were positive. The output power of the modules in the positive voltage portion was up to 30% lower than the initial power, as shown in Figure 2.6. SunPower proposed the surface polarization effect, which means that the front surface is negatively charged by the module leakage current occurring from modules in the positive voltage portion [27]. The negative charges from the leakage current are accumulated in the silicon nitride layer, since the silicon dioxide layer located below the silicon nitride has a very high resistivity, as shown in Figure 2.7. These accumulated negative charges attract positive charges, which are light-generated holes, to



the front; therefore, recombination is increased at the front. Such a high recombination rate at the surface due to the polarization effect does not require an extremely high amount of negative charges because of the exponential increase of the hole concentration with band bending [27]. It has been observed that a charge density of 1×10^{12} cm⁻² causes the band bending [27].



Figure 2.6 Module Power Degradation Distribution due to Polarization in a Series of Connected Strings Installed in an Outdoor Field [27].





Figure 2.7 Band Structure Showing Polarization [27].

It has been noticed that the polarization effect is reversible by applying opposite potential, which is negative potential to the cells with respect to the grounded frame. The blue line in Figure 2.8 shows the current–voltage (I–V) curve of the degraded module by the polarization effect. The module was recovered by applying –1000 V to the cells, with respect to the frame, for 1 hour [27].



Figure 2.8 Recovery from Polarization by Applying Opposite Potential to The Cells of Degraded Module [27].



2.3 Potential-Induced Degradation

The term potential-induced degradation (PID) was first introduced by Solon, a German solar cell and module manufacturer, in 2010 [28]. Though previously the polarization effect had been regarded as a serious problem only for high-efficiency solar cells, such as n-type back-contact solar cells or ribbon-type solar cells, the company observed an HVS degradation effect similar to the polarization effect in common p-type crystalline silicon solar cells [27]. Currently, the term PID is used widely in the PV industry for the degradation effect of standard p-type crystalline silicon solar cells. The effect of PID can be divided into three categories: system level, module level, and cell level.

2.3.1 System level

The potential difference between the cell and frame is the most important factor for PID on the system level. The system voltage depends on a number of seriesconnected modules in strings and also on temperature and irradiance. The potential difference between the cell and frame is related to the position of the series-connected modules in the string and the type of grounding. There are three potential different ground configurations: positive pole grounding, negative pole grounding, and no grounding. The potential in a string has the same polarity with positive pole grounding and negative pole grounding, as shown in Figure 2.9. When the system has no grounding, half of the modules have a positive potential and half of them have a negative potential; this is called floating potential. The floating potential is seen when a transformerless inverter that does not allow grounding is used, as shown in Figure 2.10 (b). PID is observed only from a PV system with a negative potential to the cells with respect to the



frame. In a string, modules located where there is a high negative potential are exposed to PID. There was no PID issue for p-type crystalline silicon modules when negative pole grounding (Figure 2.10 (a)) was used with a traditional transformer inverter. However, PID is becoming a serious problem on the system level since transformerless inverters often are used in PV systems due to their higher efficiency and lower cost, as compared to transformer-based inverters.



Figure 2.9 Potential Distribution for P-Type Modules in Strings Based on Type of Grounding [28].





(b) Transformerless inverter



PID-stressed modules could be characterized by taking electroluminescence (EL) images and I–V measurements. As shown in the EL image in Figure 2.11, degradation stops when the potential turns from negative to positive in the floating potential. Usually, PID can be prevented by grounding a negative pole of the string if the potential is fixed, which means not floating. However, even though the potential is fixed and properly



grounded, several ends of the modules that experience high positive potential (Figure 2.10 (a)) might experience another failure mechanism, such as electrochemical corrosion attributed to a very high positive potential in long-term outdoor operation.



Figure 2.11 EL Image of a Floating Potential String. PID is Stronger in Higher Negative Potential Portions [28].

2.3.2 Module level

It is well known that a high potential difference between the cell and frame causes a leakage current, which is composed of charge carriers (ions) [26]. The leakage current between the cell and frame in a PV module mainly depends on humidity and temperature [26, 29]. The leakage current increases as the temperature and humidity increases. The interaction of the glass, encapsulant, backsheet, and metal frame results in multiple paths of the leakage current, as shown in Figure 2.12. It was found that the leakage current flows on the surface of the glass and through its bulk to the cell in high humidity conditions, but at lower humidity, leakage current flows through the glass–encapsulant interface between the frame and cell [19, 29]. Since high leakage current in PV modules at the outdoor field was observed in very high humidity condition, the pathway relevant for PID is from the frame to the glass through the encapsulant to the cell, which is shown with a solid arrow in Figure 2.12. Other pathways shown as a dashed line in Figure 2.12 have less effect with PID [30]. Therefore, the key point in addressing PID on the module



level is to control the leakage current, which is governed by material (encapsulant or glass) properties, production processes, and module layout.



Figure 2.12 Various Leakage Current Pathways from the Metal Frame to the Encapsulated Cell [30].

One important factor in preventing PID on the module level is a resistivity of encapsulant material [28]. The encapsulant resistivity is varied by materials. For example, leakage current of encapsulant material X (low resistivity) is higher than the one of material Y (high resistivity) under the same environmental conditions, as shown in the Figure 2.13. PID could be avoided by using high resistivity encapsulant in PV module. Nowadays, EVA is used widely for encapsulant material of crystalline silicon PV modules. However, EVA is known commonly to be PID-susceptible, although PID susceptibility also varies from different EVA manufacturers, as shown in the Table 2.1. Therefore, various alternative PID-resistant encapsulant materials have been proposed to address the PID issue. One of them is silicone; however, it has shown severe degradation after 190 hours of PID testing [31]. Thermoplastic silicon elastomer (TPSE) and ionomer modules showed great resistant to PID, but the polydimethylsiloxane (PDMS) module did not protect cells from PID [32]. Even though EVA is susceptible to PID, it has been



advantageous for many reasons, such as having the best long-term stability, cost, handling, and availability. Therefore, it is preferable to prevent the PID issue at the cell level.



Figure 2.13 Leakage Current of Two Modules Having Different Encapsulant Material but Same Environmental Condition [28].

Table 2.1Four Different EVA Samples and Power before and ffter PID 85 °C/85%RH, 24 h [32]

Material type	Initial P _{max} (W)	P _{max} after PID (W)	ΔP _{max} (%)
EVA 1	0.67	0.37	-45.45
EVA 2	0.67	0.08	-88.75
EVA 3	0.67	0.07	-89.77
EVA 4	0.67	0.56	-16.84

I–V and EL also are basic PID characterization methods on the module level as well as on the system level. EL imaging clearly visualizes PID-affected cells in modules. As shown in Figure 2.14, all the cells had a good condition before PID testing, but after the PID test, several cells showed degradation as a darker area with EL.







Figure 2.14 EL Image of a Module (a) Before and (b) After PID 100 h, -1000 V [28].

2.3.3 Cell Level

SunPower found that the leakage current, which is caused by high potential with high temperature and humidity, through the front glass and encapsulation leads to accumulation of a trapped negative charge (positive charge for standard p-type base cells) on the surface [27]. Then, those accumulated charges on the front of the cells cause a breakdown of the good surface passivation, which eventually causes a power degradation. The PID effects of p-type base PV cells and modules on the cell level are similar to what SunPower has reported, but a specific mechanism causing the degradation is different since a cell type is different. It has been reported that PID is dependent on cell properties, such as base resistivity, emitter sheet resistance, and anti-reflection coating [28]. Detailed PID effect, characterization, and mechanism on the cell level are presented in the following sections.

2.3.3.1 PID and Shunting

As previously shown in the EL image of the PV modules, darker cells represent PID degradation, which leads to power loss. The PID also can be characterized by taking



I–V measurements. Figure 2.15 shows I–V curves of a typical PID stressed cell. As we can see in the figure, fill factor (FF) (2) and open circuit voltage (V_{oc}) (1) are decreased as PID progresses. The decrease of FF is attributed to a reduction of shunt resistance (R_{sh}), as shown in Table 2.2. In the cell EL image shown in Figure 2.16, shunts appear around the edge of the cell after 40 hours, and after 100 hours, the cell is completely shunted [28].



Figure 2.15 Common PID I–V Curves: (1) Shows a V_{oc} Decrease and (2) Shows an FF Decrease [28].

Table 2.2 Cell I–V Key Parameter Change during PID, –1000V, a Front Glass is flooded with Water [28]

Time (h)	V _{oc} (V)	I _{sc} (A)	P _{max} (W)	FF	$egin{array}{c} R_{ m sh} \ (\Omega) \end{array}$
0	0.615	8.24	3.616	71.4	80.4
40	0.615	8.258	3.622	71.3	51.1
80	0.6	8.109	2.658	54.6	0.5
100	0.572	7.882	1.746	38.7	0.2
relative PID	-7%	-4%	-52%	-46%	-100%





Figure 2.16 EL Image of a Cell during PID Test, -1000 V, a Front Glass is Flooded with Water [28]

Several investigations were made regarding the PID mechanism that leads to shunting of the cells. The most possible culprit for the shunting is sodium because increased sodium concentrations were observed in the surface and the emitter of PID-degraded cell by secondary ion mass spectroscopy (SIMS) [19, 33]. It was initially found that the shunts appear not as points but as planes in the busbar and finger-printed cells [34], but later Naumann *et al.* observed that the PID-shunted area has a circular shape with a diameter of 5–20 μ m [35]. The shunted areas are interrupted by busbars and fingers, as shown in the Figure 2.17. In other words, there are no shunts where busbars and fingers printed. It has been concluded that PID shunting is caused, not by metal ions from grid lines, but by positive ions, such as sodium ions from glass [34]. Since the failure mode of PID-affected cells is junction shunting, the PID also is referred to as PID of the shunting type (PID-s). Further details of the PID-shunting mechanism is presented in Section 2.3.3.5.





Figure 2.17 Dark Lock-In Thermography (DLIT) Image Showing PID Shunts. Red Areas Represent Shunts [34].

2.3.3.2 Base Resistivity

It has been observed that increasing the base resistivity of p-type cells leads to lowering PID susceptibility, as shown in Figure 2.18. The lower-base doping (higherbase resistivity) leads to a wider depletion region at the junction when the emitter doping is constant [28]. However, no further experiments or results regarding base resistivity dependence of PID have been reported yet.



Figure 2.18 PID Susceptibility as a Function of Base Resistivity [28].



2.3.3.3 Emitter Sheet Resistance

It has been reported that increasing the emitter sheet resistance leads to higher PID susceptibility [28]. The emitter sheet resistance could be increased during the cell processing. Generally, wafers are textured and then cleaned before the emitter diffusion process. If there are some residues on the front surface after the cleaning, the emitter diffusion could be affected by the residue. This might cause an increase in emitter sheet resistance so that PID increases [28]. However, another experiment shows that the dependence of the PID on sheet resistance is unclear [34]. Figure 2.19 shows the position of the cells in the module for sheet resistance. Degradation of cells in the red lines is proportional to sheet resistance, but degradation of cells in blue lines is not correlated to sheet resistance. This means that there might not be a significant correlation of PID with sheet resistance of the emitter [34].



Figure 2.19 PID Results of a Module Containing Cells with Various Emitter Sheet Resistance, EL Imaging (a) before PID and (b) after PID, (c) Emitter Sheet Resistance of Each Cells [34].



2.3.3.4 Silicon Nitride Anti-Reflection Coating

As observed by SunPower, the SiN_x ARC layer has a significant influence on the polarization effect for n-type base back contact cells [27]. Since the SiN_x ARC layer also is used widely in common p-type solar cells, characteristics of the SiN_x layer is an important key factor related to PID.

Figure 2.20 shows the dependence of PID on the silicon–nitrogen ratio of the ARC, which corresponds to the refractive index (RI). There are three findings regarding the correlation between the ARC and PID from the figure. First, silicon-rich layers (high RI) tend to have lower degradation than nitrogen-rich layers (low RI). Second, in addition to the RI, PID also is affected by the layer thickness. The degradation of a thinner ARC layer is slightly lower than one that has a thicker ARC layer. Therefore, a thinner ARC layer with a higher silicon–nitrogen ratio makes it possible to design less PID-affected cells. Third, PID is affected by the ARC deposition process, such as process B in Figure 2.20 [28]. Those processes have not been disclosed by the authors yet.



Figure 2.20 PID Susceptibility as a Function of RI, Thickness, and Deposition Process [28].



Figure 2.21 (a) shows the shunt conductance measured after the PID test as a function of the RI for three different deposition processes [36]. The shunt conductance (PID-susceptibility) is reduced with an increasing RI [36], which is consistent with the previous experiment [28]. Thus PID could be prevented by increasing RI. However, it is not recommended to increase the RI higher than 2.1 because the parasitic light absorption in the ARC also increases [36]. Thus, multiple SiN_x layer experiments were performed by Koch et al. in order to minimize PID as well as reflectance loss [37]. The double-layer ARCs are one with a standard RI of 2.08 and one with a varying RI between 2.2 and 2.5. A triple layer ARC with a similar combination to the double-layer ARC also was used for the experiment. However, it showed that those multiple ARC layers do not have a large impact in preventing PID, as shown in Figure 2.22. As reported in [28], the SiN_x deposition equipment has an impact on PID. In the case of using the standard SiN_x layers with an RI of 2.1, cells with a SiN_x ARC processed by equipment A showed about 5.5% efficiency degradation, whereas no efficiency degradation was observed from the ones processed by equipment C [36]. Therefore, it is also important to choose appropriate SiN_x deposition equipment in order to fabricate PID free cells [36]. Unfortunately, neither groups disclosed what process or equipment was used. Figure 2.21 (b) shows the results of an accelerated PID stress test (90 °C, 6-day, -2000 V), for ARC-modified cells. Less than 0.5% power degradation was observed for the modified cells, although common EVA was used. Therefore, the modified SiN_x ARC effectively could protect the cells against PID [36].





Figure 2.21 (a) Measured Shunt Conductance $(1/R_p)$ as a Function of SiN_x RI with Different Equipment. (b) Measured Normalized P_{max} Degradation from Various PID Modified Modules [36].



Figure 2.22 PID Degradation from Various ARC Layer Structures. A: Single Layer, DL: Double Layer, and M, P: Triple Layer [37].

2.3.3.5 PID Mechanism

As presented in previous sections, negative potential, with respect to ground, leads to leakage current, which results in positive-charge accumulation on the cell surface in p-type crystalline silicon modules. For thin-film modules, it has been observed that



negative potential, with respect to ground, on the active layer can pull sodium ions from soda-lime glass [38]. Sodium ions also were observed in crystalline silicon solar cells by SIMS after PID testing, and it has been shown that the PID cell has a locally shunted region, which is characterized by EL and lock-in thermography (LIT) [19, 39, 40]. On the shunted region, disappearance of the electron beam induced current (EBIC) signal in the p-n junction of the PID cell was observed by Naumann *et al.* [39], and Bauer *et al.* observed a high sodium ion concentration in the locally shunted region, as shown in Figure 2.23 (d) [40].



Figure 2.23 (a) Lock-In EBIC Image of the PID-Stressed Cell; (b) Lock-In EBIC Image of the Non-PID-Stressed Cell; (c) Detailed Lock-In EBIC Image of the PID-Stressed Cell; (d) SIMS-Difference Na⁺ Map of Red Area in (c) [40]

According to [39], the sodium ions, which might be from soda-lime glass, could be diffused to the SiN_x ARC through the encapsulant material by huge potential, and then they are accumulated at SiN_x/Si interface. It was assumed that the sodium ions attract negative charges in the n⁺ emitter layer, and the negative charges result in the repulsion of majority charges in the emitter surface. Therefore, band bending could occur in the



emitter region, as shown in Figure 2.24 (b), and finally, the built-in field gradient at the p–n junction would disappear [39]. Bauer *et al.* [40] proposed the emitter inversion model as a PID mechanism based on the results in [39]. According to their proposed model, positive charges (Na⁺) attract or create negative charges, and then the ionic double layer is formed in the SiN_x layer, as shown in Figure 2.24 (a) [40]. Electric fields are formed on both side edges of the ionic double layer, and they repel the electrons in the n⁺ emitter layer. This situation causes local emitter inversion, which is from n⁺ to p⁺. The tunneling current due to the new p⁺–n⁺ junction leads to the shunting. Bauer *et al.* presented that the amount of negative surface charge to make the inversion is -1×10^{15} cm⁻³ [40]. This is a huge amount of charges, and it may not happen in a real situation. Also, the nature of negative charges is not clear in this proposed PID model [40].



Figure 2.24 (a) Proposed Emitter Inversion Model [40]; (b) Energy Band Diagram Showing Surface-Charge-Induced Band Bending, before PID (Solid) and after PID (Dashed) [39].



Currently, the most promising PID mechanism is explained by stacking faults [35, 41-43]. Naumann et al. have shown that the dark area in an EL image is identical to the high-temperature region in a DLIT image, which can visualize shunting, and the presence of sodium in that region was confirmed by time-of-flight SIMS (ToF-SIMS) [41]. The cross section of the PID-affected region has been investigated by transmission electron microscopy (TEM) and energy dispersive X-ray spectroscopy (EDX) mapping, and a stacking fault has been observed in the region [42]. Figure 2.25 shows that there is sodium in the stacking fault region as well as the SiO_x layer. The length of the stacking fault has been known to be about 3 μ m from the surface of the cell [42]. It has been observed that the stacking fault is decorated and rearranged by sodium, as shown in Figure 2.26. According to [35], the original defect might be an intrinsic stacking fault (0.32 nm), because the width of the sodium-decorated stacking fault (0.57 nm) is smaller than the extrinsic stacking fault (0.63 nm). The source of the sodium causing PID is unclear. It has been speculated that the sodium is from soda-lime glass [44] or the SiN_x layer as a contaminant [45].





Figure 2.25 (a) TEM Image Showing the Stacking Fault of a PID-Stressed Cell, (b) EDX Intensity Maps Show Sodium Accumulation in a SiN_x/Si Interface and Stacking Fault Area, (c) High-Resolution TEM (HRTEM) Images of the Stacking Fault [42].



Figure 2.26 (a) Brightness (intensity) and Dimensions of a PID-Affected Stacking Fault; (b) The Corresponding High-Angle Annular Dark Field (HAADF) TEM Image Shows Atomic Structure of the PID-Affected Stacking Fault [35].

According to [35], sodium ions are driven toward the SiN_x/Si interface under an

electric potential difference less than 0.05 V over the SiN_x layer. As shown in Figure 2.27



(a), the sodium ions are accumulated on the SiO_x layer, and then, they stop additional sodium transport from the surface to the SiO_x layer. The sodium ions get into the stacking fault and start being neutralized by free electrons in the emitter of the cell. Therefore, more sodium ions could move to the SiO_x layer from the SiN_x layer. The sodiumdecorated stacking fault could be converted to a quasi-metallic layer, which is represented by a continuum of electronic states around the Fermi level, therefore ohmic conductivity across the p-n junction causes shunting [35]. A detailed explanation based on the band structure is shown in Figure 2.27 (b). The sodium atoms in the stacking fault widen (~ 0.25 nm) the initial intrinsic stacking fault, and Naumann *et al.* assumed that concentrated sodium atoms at the stacking fault cause a band of gap states at multiple energies within the Si band gap, as shown in Figure 2.27 (b) [35]. This mechanism depends on the local defect-level concentration. Processes 1 and 2 in Figure 2.27 (b) represent a high local defect-level concentration and a low local defect level concentration, respectively. In process 1, the defect orbitals of neighbored levels overlap, therefore hopping conduction occurs. Thermal activation is not required for this process; therefore, ohmic shunting across p-n junction appears [35]. In process 2, which is thermally activated, these defect levels, which are not enough to enable the hopping conduction, lead to an increase of depletion region recombination (J_{02}) with a large ideality factor (n_2) [35, 43]. The increase of J_{02} and n_2 after PID stress has been observed in [33, 43, 46]. This proposed PID mechanism based on the stacking fault has been investigated by density functional theory and was supported by the calculated results [47].





Figure 2.27 (a) Cross Section of Solar Cell and (b) Band Diagram of the Proposed PID Mechanism Using the Stacking Fault [35].

PID has been known for reversible degradation; therefore, the PID recovery mechanism also was investigated. In recent study [48], sodium removal from the stacking fault confirmed by scanning transmission electron microscope (STEM)/EDX was observed after the thermal recovery process (250 °C) as shown in Figure 2.28. The localized shunting disappeared due to sodium out-diffusion from the stacking fault by thermal recovery, and this was confirmed by I–V and EBIC [48]. The band structure of the recovery mechanism proposed by [48] is shown in Figure 2.29.





Figure 2.28 HAADF STEM Images, (a) before Thermal Recovery (after PID) and (b) after Thermal Recovery [48].



Figure 2.29 Band Diagram Showing (a) Proposed PID Mechanism and (b) PID Recovery Mechanism [48].

2.4 PID Test Methods

PID has become an urgent reliability issue due to its rapid and strong P_{max} degradation. One of the concerns regarding PID in the commercial PV module industry is that PID cannot be examined by the crystalline silicon terrestrial design and qualification test, International Electrotechnical Commission (IEC) 61215 Ed. 2 standard [49]. Thus,



the IEC 62804 draft standard (System voltage durability test for crystalline silicon modules - Qualification and type approval) is being developed [50]. Several test methods have been developed for PID stress on the module and cell levels.

2.4.1 Chamber Method

The chamber test method uses an environmental chamber to simulate PID conditions observed from the field. The IEC 62804 draft standard sets minimal PID stress levels [51]. The stress condition for detection of PID in the chamber is 60 °C \pm 2 °C and 85% \pm 3% RH with a 96 h dwell time. Applied voltage is dependent on module rated system voltage and polarities. Typically, negative potential (-600 Vdc or -1000 Vdc) is applied to shorted leads of PV modules (p-type c-Si modules), as shown in Figure 2.30. A leakage current can be monitored by a voltmeter across a resistor, such as an R1 resistor, shown in Figure 2.30. The whole module surface, including the aluminum frame, will be conductive due to the conditions of 60 °C/85% RH; therefore, the leakage current causing PID is generated between the glass surface and the aluminum frame due to the potential difference (up to -1000 Vdc).



Figure 2.30 A Schematic Showing a PID Setup for the Chamber Method [33].





2.4.2 Metal Foil Method

Instead of depending on 85% RH, the metal foil method uses a sheet of aluminum or copper foil as a conductive layer to make a front surface, including the aluminum frame, conductive [46, 52-55]. This method does not need the environmental chamber since the metal foil provides uniform contact all over the surface and frame. The test temperature and time are 25 °C and 168 h, respectively. The module rated system voltage is applied to a test module. This simple PID test method may cost less than the chamber test method, but it takes longer to detect reasonable PID. Soiling on the PV module glass may increase PID susceptibility [56], and the metal foil test condition could represent PID of heavily soiled modules [57]. Therefore, the metal foil method could be used for soiled-PV module PID testing research.



Figure 2.31 Schematic Drawing of (a) the Metal Foil Method and (b) a Photograph [58].

2.4.3 Corona Discharge Method

This is a cell-level PID testing method that does not require module lamination. Instead of sodium ions (Na⁺), positive charges are deposited intentionally on the sample



surface with high voltage (~ 11 kV) [30, 34]. Therefore, the tested cell showed PID degradation. This method is quick and convenient in carrying out PID testing. The corona discharge setup is shown in Figure 2.32.



Figure 2.32 Corona Discharge Method for PID Testing [34].

2.5 PID Solutions

Although the PID mechanism is still unclear, there are various methods for avoiding PID. PID can be avoided on three different levels: cell level, module level, and system level.

2.5.1 Cell Level

The most well-known method for avoiding PID is the modification of the SiN_x ARC layer. PID can be suppressed by increasing the RI of the SiN_x ARC film [28, 34, 36, 37, 59]. Increasing the RI of SiN_x ARC (> 2.1) should be considered carefully because the cell efficiency will be decreased as the RI increases. Therefore, compromise is needed to avoid PID as well as efficiency loss in ARC. It also was reported that the deposition process of the SiN_x ARC could make a change in PID susceptibility [28, 60, 61]. However, the detailed information of the different processes for the SiN_x deposition has not been disclosed.



Another method to prevent PID is adding a thin oxide layer (SiO₂) between the SiN_x layer and the emitter layer of the cell [62-65]. The SiO₂ layer could be added during post implant anneal when the ion implantation process is used to create the emitter. For common POCl₃-diffused solar cells, the thermal oxidation process typical for oxide growth also is used to grow the SiO₂ layer. A UV lamp also could be used to grow the oxide layer [65]. A suitable thickness for the oxide layer to prevent PID has been found to be 7–10 nm [63]. Figure 2.33 shows that the SiO₂ layer is comparable for the two different oxidation processes.



Figure 2.33 SEM Image of a Solar Cell Cross Section of (a) an Ion Implanted Cell and (b) a Modified POCl₃ Sample, Which has a Diffusion-Furnace-Grown SiO₂ after Phosphosilicate Glass (PSG) Removal [63].

Recently, it was observed that a well-controlled phosphosilicate glass (PSG) layer between the SiN_x ARC and the emitter protects cells from PID [66]. The SIMS profile of this PSG-layer cell showed a decrease in sodium where the PSG layer is located, as shown in Figure 2.34. The specific mechanism that blocks PID in the PSG layer is unclear. It is speculated that the PSG layer acts as a sodium-gettering layer [67].





Figure 2.34 SIMS Depth Profile of (a) a Standard Solar Cell and (b) the PSG Layer Added to the Solar Cell [66].

2.5.2 Module Level

Even if PID-susceptible cells are used in building the PV module, PID still could be prevented by using alternative materials on the module level. It has been observed that the amount of leakage current is related to PID susceptibility, although it is not always proportional [28]. Therefore, PID can be suppressed by using high volume resistivity encapsulant materials (~ $7 \times 10^{16} \Omega$ -cm) hindering the ionic current flow through the



encapsulant instead of using EVA (~ $1 \times 10^{14} \Omega$ -cm), which is one of most common encapsulant materials in commercial PV modules. Those encapsulants include ionomer [68, 69], polyolefin [70], and PID-resistant EVA [71]. As shown in Table 2.3, EVA is one of the low volume resistivity materials, along with PVB, so these materials are not a good choice for building a PV module with PID-susceptible cells. Another method to block the ionic current flow through the cell is adding additional material between the module glass and the cell. It has been shown that a thin film of polyethylene (~ 30 µm thickness with volume resistivity material of $1.8 \times 10^{17} \Omega$ -cm) between the glass and PID-susceptible EVA effectively prevents PID [72]. A Corning Willow glass sheet added between the PV module glass and the cell also perfectly prevents PID. A detailed explanation is presented in Chapter 7.

Encapsulant material	Volume resistivity (Ω-cm)	Transmittance (%)
EVA	1.0×10^{14}	91.0
PVB	4.8×10^{12}	91.0
TPU	2.7×10^{14}	90.0
Silicone	6.0×10^{15}	98.9
Polyolefin	2.0×10^{16}	92.0
Ionomer	7.1×10^{16}	93.4

Table 2.3Electrical and Optical Properties of Different Encapsulant Materials [70]



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Using alternative PV module glass is another way to avoid PID on the module level. Soda-lime glass that is PID-susceptible has been used widely as a common PV module glass. It has been reported that PID was suppressed when quartz or borosilicate glass was used as a PV module glass [19, 33]. A PV module with an aluminosilicate chemically strengthened glass instead of soda-lime glass showed strong PID resistance, although its thickness is only about 0.85 mm [73]. Another candidate is an acrylic-film material, which is used widely for outdoor structures, such as signboards. It has been shown that the PV module composed of an acrylic-film coversheet instead of the soda-lime glass was reported as a PID-prevention method. TiO₂ film coated with a thickness of about 200 nm inside of PV module glass showed great PID resistance [75]. Hydrophobic-ARC-coated PV module glass also showed a reduction of PID [76]. Chemical coating above the PV module glass is another way to avoid PID [77].

2.5.3 System Level

One interesting characteristic of PID is that the PID degradation is reversible. PID-affected cells and modules can be recovered by applying reverse potential [28]. The PV inverter company SMA has been offering a system-level PID solution, which is called PV Offset Box [78]. The PV Offset Box is an accessory that works with a PV inverter and enables reverse potential for the PV modules at nighttime for PID recovery. Therefore, PID modules that are affected during the daytime could be recovered at night.

Using a micro inverter also avoids PID on the system level [36]. Most micro inverters are rated to match one PV module. Therefore, DC voltage of the entire PV system using micro inverters is less than 50 Vdc, which is classified as extra-low voltage,



while the common PV system using a normal inverter is up to 1000 Vdc. Since the system voltage using micro inverters is extremely low, there would not be a chance for PID.

2.6 PID of Other Cell Technologies

As previously mentioned in Section 2.3, PID-s (PID of the shunting type) is only observed from p-type crystalline silicon solar cells. Since PID became a serious issue for p-type cells, different technology cell and module manufacturers carried out PID testing to ensure the reliability of their products. The heterojunction with intrinsic thin-layer (HIT) solar cells have been known to be PID-resistant regardless of applied bias [79, 80]. The HIT cell is basically an n-type cell that uses transparent conductive oxide (TCO) as an ARC [81], while common p-type diffused junction solar cells use a SiN_x ARC [82]. It is thought that TCO, which causes no charge accumulation on the cell surface, is a much more conductive layer than the SiN_x ARC, which is an insulating layer that causes charge accumulation, and makes the HIT cell PID-resistant [83]. Arizona State University (ASU) Solar Power Laboratory (SPL) also has an ability to fabricate HIT cells in a pilot line. The SPL single HIT cell coupon was tested for PID and showed PID-resistance regardless of polarity of applied bias, as shown in Figure 2.36.



Figure 2.35 HIT Cell Structure [84].






Figure 2.36 PID Test Results of ASU SPL HIT Cell.

Another common n-type cell is an interdigitated back contact (IBC) solar cell, such as the SunPower cell. Naumann *et al.* [85] investigated the PID effect of IBC solar cells. They observed a degradation of a front side passivation layer from the IBC cell in contrast to the p-type cell, which shows junction shunting. A loss of the passivation layer causes an increase of surface recombination, which results in current loss in the cell operating condition. The mechanism is different from PID-s, so it is called a degradation of the front side passivation layer (PID-p) [85]. Interestingly, the PID result of their n-type IBC cell showed the degradation when positive voltage was applied [85]. This result was the opposite with SunPower's IBC cell, which was known to be sensitive to the polarization effect. However, the n-type IBC cell with the front floating emitter showed the PID-p when negative potential was applied [86], while the n-type IBC cell with the front surface field [85] experienced the PID-p at positive potential.



The PID-p also was observed from the n-type front junction cells with negative potential applied [87]. Due to the negative voltage to the n-type front junction cell, positive charges caused by the leakage current are trapped in the SiN_x, as shown in Figure 2.37 (b). The trapped positive charges lead to an increase in the minority carriers (electrons) in the p-type front layer; therefore, the surface charge recombination increases. Hara *et al.* [87] observed that the n-type front junction cell experienced PID-p even when chemically strengthened glass (CSG) [73], known as an alternative glass material for p-type modules for preventing PID-s due to its very low sodium concentration, as mentioned in Section 2.3.2, was used for the module glass. Therefore, it was concluded that PID-p for an n-type front junction cell is not related to the presence of sodium in glass.



Figure 2.37 Schematic Diagram Shows the Proposed Mechanism of PID-p in N-Type Si PV Modules. (a) N-Type IBC Cell Reported in [27]. (b) N-Type Front Junction Cell [87].



Fjällström *et al.* [88] investigated PID of copper-indium-gallium-selenide (CIGS) thin-film solar cells. They showed that sodium concentration in the module glass strongly affects PID susceptibility of CIGS modules, as shown in Figure 2.38. Recovery from PID also was observed both under reverse potential with high temperature and no potential with room temperature. The CIGS cell with a CdS buffer layer clearly showed the recovery while the one with a Zn(O,S) buffer layer showed no recovery [89]. The type of glass and the type of buffer layer in CIGS modules play an important role in PID susceptibility, but the mechanism is not fully understood yet [88, 89].



Figure 2.38 PID Susceptibility Depending on the Type of Glass Including Soda-Lime Glass (SLG) Used in the CIGS Modules [88].



Chapter 3

POTENTIAL-INDUCED DEGRADATION AND INCOMPLETE RECOVERY

3.1 Introduction

One of the high-voltage stress (HVS) based performance-loss issues experienced by photovoltaic (PV) modules in the field is called potential-induced degradation (PID) [28]. It has been observed that PID-stressed PV modules showed significant output power decrease ($\sim 30\%$), even more at low irradiance level, and increase of hot-spot risk [28, 37, 90]. In response to this field issue, a new IEC standard for PID is being developed [50, 91]. This degradation is mainly due to a cell junction shunting [33, 35, 36, 43], so the PID is more specifically called PID of the shunting type (PID-s) [35, 43] in order to discriminate other PID, such as electrochemical corrosion [17]. Sodium has been suspected to cause PID [33], and source of sodium identified in the PV module glass [19], which is usually soda-lime glass. It has been shown that sodium decorated stacking faults located from the cell surface through junction lead to shunting through a thin quasimetallic layer [35, 42, 43]. The PID/PID-s has only been observed with p-base crystalline silicon (c-Si) cells negatively biased to the frame. This issue could technically be addressed at the cell/module manufacturing level or at the installed system level. At the manufacturing level, the PID issue can be addressed by modifying the silicon nitride antireflection coating (ARC) [28, 37], by preventing ion conduction through encapsulant or by eliminating sodium content in the glass superstrate [19, 32]. If the system is already installed in the field with PID susceptible cells/modules, the PID issue can still be addressed by applying a reverse potential on the modules during the nighttime [36, 52] so



that the cells are positively biased to the frame. Since the energy consumption during this recovery period is so small when compared with the energy production during the daytime, the reverse-potential approach to address the PID issue has been implemented by many system owners. However, recent papers separately discuss an incomplete recovery under reverse potential in terms of efficiency at high and low irradiance levels, shunt resistance of cells, and the quantum efficiency (QE) [33, 52, 92]. Moreover, QE measurement on PID-stressed cells has not been thoroughly investigated yet, although QE loss at short wavelength was observed and briefly reported in PID-stressed cells by our research group [92, 93]. QE curves of solar cells are obtained by measuring the cell's light-generated current at various wavelengths when monochromatic light shines on the cell. This measured current is very sensitive to the cell's shunt resistance, and it may lead to scaling error when shunt resistance is very low in, for example, PID-stressed cells. In this chapter, we present a detailed analysis on the effectiveness of the reverse-potential method on the recovery of the PID-subjected cells and incompleteness of QE recovery, which have been previously published [92, 93]. This chapter also presents the challenges in measuring accurate QE of heavily shunted cells and the QE results obtained using a newly developed ultra-low impedance accessory. Investigation of the source of sodium is presented in last section of this chapter.

3.2 Experiments

Test coupons (one-cell laminates) were constructed using common commercialgrade construction materials of glass, EVA, cell, and backsheet. The experiments were performed using 156 mm × 156 mm size p-base monocrystalline silicon cells, which are susceptible to the PID issue. The PID stress experiments were carried out at -600 V for



88 h at two different temperatures (60 °C and 85 °C) with 0% relative humidity (RH). The front glass was fully covered with an adhesive conductive aluminum tape to obtain uniform conductivity throughout the glass surface. The negative voltage was applied to the shorted leads of test cell-coupon and the positive voltage was applied on the aluminum tape covering the front glass surface.

Before and after the PID test, all test samples were characterized by light I–V (LIV), dark I–V (DIV), electroluminescence imaging (EL), infrared imaging (IR), and quantum efficiency (QE). A steady-state solar simulator was used to take the I–V measurements. The recovery experiments were carried out using a reverse potential of +600 V for 88 h at 60 °C, 0% RH. The reverse-potential-subjected samples were characterized using the techniques identified above and then stored at room temperature without any imposed potential and periodically characterized to observe additional recovery, if any. Also, some of the PID-stressed samples that were not subjected to reverse-potential recovery were stored and periodically characterized at room temperature to compare the recovery rates between the reverse-potential samples and room temperature stored samples. The recovery rate and extent are reported in terms of: power recovery at both high and low irradiances; shunt resistance recovery, and quantum-efficiency recovery at wavelengths between 350 and 1100 nm.

The QE measurement artifacts of heavily shunted cells can be simulated using an externally connected parallel resistors (R_p), as shown in Figure 3.1. If the QE system input impedance is not exactly zero, a small voltage drop will develop at the QE system connection, driving current through the external R_p instead of to the QE system to be measured. Thus, the measured QE decreases. This wavelength-independent scaling error



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in QE can be significantly reduced by utilizing a very-low impedance method. A new ultra-low impedance accessory developed by PV Measurements, Inc. has been utilized in the QE system of this work to minimize the QE drops of the heavily shunted cells.



Figure 3.1 Basic Circuit Diagram of a Solar Cell Indicating Shunt $(R_{sh} \text{ or } R_p)$ and Input Impedance of QE System

3.3 Incomplete Recovery of Power and Shunt Resistance

Figure 3.2 (a) shows the progress of PID and recovery of maximum power (P_{max}) at 1000 W/m² and on shunt resistance. After 88 h of PID stress at 60 °C, the remaining power was determined to be 70% when compared with the initial power, and the shunt resistance dropped to under 1 Ω . After the PID test, the sample was stored at room temperature with no potential for more than 75 days, and the power at 1000 W/m² is determined to be recovered to about 92%. However, the recovery of shunt resistance is still very low although most of the power at this high irradiance level is recovered.

At 85 °C PID stress, a much higher power drop in a shorter time of test was observed, as shown in Figure 3.2 (b). After leaving at room temperature for more than 500 h, the recovered power was only 87% at 1000 W/m² irradiance level and 60% at 240 W/m² irradiance level, and the recovered shunt resistance is practically insignificant.



Also, it takes much longer time for both P_{max} and shunt resistance to be recovered when compared with the 60 °C-PID-stressed samples due to extensive damage to the cell materials and/or junction, which may not be a realistic observation in the field.



Figure 3.2 (a) Recovery Rate and Level of Output Power at High (1000 W/m²) Irradiance Level, and Shunt Resistance during the Recovery Period at Room Temperature (No Aluminum and No Bias for 1992 h) on a Test Sample Subjected to the PID Stress at 60 °C/0% RH (Aluminum Covered, -600 V, 88 h). (b) Recovery Rate and Level of Output Power at High (1000 W/m²) and Low (240 W/m²) Irradiance Levels, and of Shunt Resistance during the Recovery Period at Room Temperature (No Aluminum and No Bias for 576 h) on a Test Sample Subjected to the PID Stress at 85 °C/0% RH (Aluminum Covered, -600 V, 44 h).

To observe if an applied reverse voltage at a higher temperature than the room temperature could recover the cells to a higher level, another set of test coupons were prepared and subjected to the tests, as shown in Figure 3.3. There is practically no difference in recovery of power at 1000 W/m² irradiance level, between 0 V and +600 V potential, but the reverse potential at high temperature accelerates recovery from PID in a shorter amount of time compared to room temperature storage with no voltage



application as described in [52]. Even though the recovery of power at high irradiance of 1000 W/m² is as high as 93% of its original, the recovery of the shunt resistance is still only about 40% of its original. Neither of those cells showed higher than 50% shunt-resistance recovery after 125-days of room temperature storage. And, recovery speed for both power and shunt resistance is extremely slow after 40-day storage. This poor recovery of shunt resistance would have a serious impact on the cell efficiency at low irradiance levels [90].



Figure 3.3 Normalized Power and Shunt Resistance. Both Coupons Have the Same PID Conditions (60 °C, -600 V, 88 h) but Different Recovery Methods. Coupon A: 15-Day Room Temperature Storage with No Bias, Coupon B: +600 V @ 60 °C, 88 h.

This inadequate shunt resistance recovery could cause: (i) a safety issue if the cells in a module were to operate under shaded/reverse-bias condition with failed bypass diodes; (ii) very-low energy production at the sites where the module performance



primarily depends on the prevailing low-light conditions. All the fresh test coupons showed practically 0 A current flowing through the cell at -12 V. Due to cell shunting after the PID stress, the observed reverse current at -9 V was higher than 8 A just after the stress test and even after a 24-day recovery period at room temperature. Because of this localized high current in reverse bias, as shown in Figure 3.4 (b) and (c), the cell was found to be damaged due to high localized temperature, as shown in Figure 3.4 (d). The light emission (white spots) shown in the reverse bias EL image (Figure 3.4 (b), is attributed to different type of shunts (ohmic or non-ohmic), in this case shunts are caused by PID, and source of shunt can be identified by spatially resolved EL imaging with lockin thermography (LIT) under reverse and forward bias [94]. It has been reported that light emission in EL under reverse bias is also related to electrical breakdown [95]. Interestingly, it was observed that this cell was not permanently damaged, i.e., breakdown. The damaged cell was stored at room temperature for 32 days and then I-V and EL were carried out. The results, as shown in Figure 3.4 (e), show that there was increase of P_{max} and shunt resistance when compared with the day that cell was damaged due to reverse-bias EL measurement. EL images (not shown here) also showed brighter damaged area than before. The cell has been monitored more than 200 days to see if there is complete recovery from PID; however, no complete recovery has been obtained. This result supports that the remaining sodium atoms in the stacking faults or the remaining sodium ions in other regions, such as SiN_x or SiO_x of the cell [48] still have a critical impact on PID-stressed cells. It is assumed that those remaining sodium atoms hinder the 100% recovery of shunt resistance and cause a very-high possibility that PID-stressed cells could be easily damaged under reverse-bias condition.





Figure 3.4 EL and IR Images of 24-day RT Stored PID-Stressed (85 °C, 44 h) Cell. (a) Forward-Bias EL, (b) Reverse-Bias EL, (c) Reverse-Biased IR Image, (d) Forward-Bias EL after Cell Damaged by Reverse Bias in Step b. (e) The Recovery Method Applied for These Results is the Room Temperature Storage with No Voltage Application. The Reverse Bias Imposed during EL Decreased the Shunt Resistance. Initial R_{sh} was 183 Ω .

Even the coupon with 92% power recovery (after 83 days at room temperature) showed a reverse current of 0.4 A at -7 V and this localized current could be high enough in damaging the cell under reverse-bias condition. To further investigate the PID effect in



terms of light I–V in negative voltage, another coupon was built and PID stressed at 60 °C/–600 V/88 h. Figure 3.5 shows I–V curves of the coupon at 240 W/m² irradiance level, which is a worse-performance condition for a shunted cell. There was a large reverse current after 88 h PID while a fresh cell had no reverse current at all, as shown in Figure 3.5. The 98-h PID recovery of the cell (94% P_{max} at 1000 W/m² and 75% P_{max} at low irradiance, respectively) still showed high reverse current due to very slow recovery of R_{sh}. Therefore, it is suggested that R_{sh} should be monitored with P_{max} in evaluating PID recovery. Additionally, it was observed that the increase in reverse current is nonlinear, which represents that shunting caused by PID is not a simple shunting. The presence of localized shunts even after extensive recovery with or without reverse potential indicates the presence of traces of sodium at the junction.



Figure 3.5 I–V Characteristics of PID-Stressed and Recovered Cell at Low Irradiance (240 W/m²) (PID 60 °C/–600 V/88 h; PID Recovery at Room Temperature/No Bias/98 h).



3.4 Scaling Error of Heavily Shunted Cells

The QE measurement becomes very sensitive in the heavily shunted cells. If the series resistance of the test cell along with the input impedance of the QE system is substantial, then a significant voltage develops at the point of current generation. That voltage then drives current through the shunt in the cell, allowing less current to exit the solar cell and enter the QE system. This voltage significantly diminishes the measured QE, especially in the heavily shunted cells. The measured QE of the shunted cells will only shift up or down (scaling error) with no wavelength-dependent error. Figure 3.6 (a) shows QE curves, obtained at various PID recovery stages, of the same test coupon used in Figure 3.2 (b). It was initially thought that the large QE drop shown in Figure 3.6 (a) was due purely to the PID effect of the test device with no influence from the measuring equipment. When the integrated short circuit current (I_{sc}) from the QE curve was compared with the measured Isc from the I-V curve, it was realized that there is a significant influence of test equipment on the accuracy of the QE curve. The decrease of I_{sc} based on the integration of the QE curve after PID is about 55%, which is a great difference than the I_{sc} from the conventional white light I–V measurement (7%).





Figure 3.6 (a) QE Curves of Same Test Coupon as Shown in Figure 3.3, (b) QE Curves of a Fresh Cell Coupon Connected with Various Parallel Resistors (R_p) .

Since there were experimental complications in controlling the shunt resistance with high level of repeatability, a physical resistor was used to perform the simulated-PID



experiments. To simulate a PID shunted cell, an external resistor was connected in parallel across the two leads of a fresh coupon that was not subjected to any PID stress. As shown in Figure 3.6 (b), QE with various parallel resistances (R_p) followed the same trends as that of the PID-affected cell. This result supports that the QE drop in a PID-affected cell is not due purely to the PID effect but also partly to the measuring equipment impedance.

To measure the QE of heavily shunted cell accurately, QE curves of the cell could be normalized and the scaling problem could be alleviated by matching its integrated I_{sc} with the white light Isc of I-V measurements. Another way to address this scaling problem is to make the input impedance of the QE equipment as small as possible. Therefore, an accessory having very-low input impedance, developed by PV Measurements, Inc. was connected to the existing QE system. Figure 3.7 shows the measured QE curves obtained without and with low impedance accessory; this figure also includes the QE curves obtained with three 1- Ω resistors incrementally connected in parallel. Encouragingly, the QE curves (with parallel resistors) with the low impedance accessory have dramatic improvements (Figure 3.7 (b)) when compared with the ones without it (Figure 3.7 (a)). Therefore, it is highly recommended that QE measurements on PID-affected cell use a QE system with low input impedance. Input impedance of QE system without the very-low input impedance accessory was measured at 0.35 Ω while the one with very-low input impedance accessory is about 0.015 Ω . Measuring a severely shunted cell, (e.g. $R_{sh} = 1 \Omega$), needs an input impedance smaller than 0.015 Ω in order to minimize scaling error. The use of low impedance accessory is determined to be a must for the nonintrusive QE measurements of heavily shunted PID cells connected in series in



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a commercial PV module [96]. The use of pulsed light bias (PLB) or pulsed voltage bias (PVB) method for the multijunction solar cells with low shunt resistances [97] may need to be investigated to further improve the accuracy of the existing QE system, which is already installed with the low impedance accessory.



Figure 3.7 Measured QE Curves of a Fresh Cell Coupon ((a) and (b)), PID Cell Coupon (c) and Shunted Cell (d)): (a) without Low Impedance Accessory, (b) with Low Impedance Accessory (Three 1- Ω Resistors were Incrementally Connected in Parallel to the One-Cell Coupon Leads Shown in (a) and (b)). (c) Effect of Low Impedance Accessory on PID-Affected One-Cell Coupon, (d) Effect of Low Impedance Accessory on Heavily Shunted Cell.



3.5 Incomplete Recovery of Quantum Efficiency

Figure 3.8 (a) shows QE curves of coupon B, identified in Figure 3.3, after PID (P_{max}: 81%) and reverse-potential recovery (P_{max}: 96%) steps. The QE of after-88 hrecovery stressed/shunted sample was found to be much lower than the initial due to a reason mentioned in Section 3.4. To observe if the relative QE curves between the fresh/initial and recovered cell are identical, these curves were normalized. As shown in Figure 3.8 (a), the QE values for the photons between 350 and 600 nm wavelength are found to be lower than the QE of fresh/initial cell. This phenomenon was clearer in 85 °C PID-stressed cell, as shown in Figure 3.8 (b). None of the QE curves in recovery process were identical to initial ones at wavelength range from 350 to 600 nm. QE drop at short wavelength range shows that there is another PID effect in addition to junction shunting since QE measurement of shunt-only cell affects the overall QE scale down as mentioned in Section 3.4. It is well known that change of QE at short wavelength range is caused by changes in front surface recombination velocity or emitter diffusion length [98]. Further details are described in Chapter 4. It was shown that a huge increase in the front surface recombination velocity is necessary in aligning the PID mechanism to the inversion model [99]. In addition, a degradation of front surface passivation that leads to an increase in surface recombination was observed in interdigitated back contact (IBC) solar cells after PID stressing [85]. Therefore, PID stressed cells might possibly have a high front surface recombination velocity. The mechanism for front surface recombination velocity increase after PID is not fully understood and it will be a subject of future research.





Figure 3.8 (a) QE of Coupon Used in Figure 3.3, QE of Fresh/Initial, PID-Stressed and Recovered Cells (PID Stress at 60 °C/–600 V/88 h; PID Recovery at 60 °C/+600 V/88 h), (b) Normalized QE of Figure 3.6 (a), QE of Fresh/Initial, PID-Stressed and Recovered Cells (PID Stress at 85 °C/–600 V/44 h; PID Recovery at RT/No Bias/408 h).



3.6 Effect of Sodium Concentration in Glass on PID

The transport of sodium from the module glass to a cell surface is strongly suspected to cause PID. Using an alternative glass with no sodium content, such as quartz, effectively protects cells in a module from PID [19, 73]. However, it is not clear that a particular amount of sodium is critical for PID. In order to further investigate the effect of sodium content in the glass on PID, two one-cell coupons using different type of glass were prepared and stressed in a condition described in Section 3.2. The only difference is that 60 °C/85% RH was applied to the test conditions instead of using front aluminum covered tape. The encapsulated cells were standard commercial cells known to be susceptible to PID. Coupon 1 used soda-lime glass as is typical in commercial PV modules. Coupon 2 used borosilicate glass, which has a lower sodium content than sodalime glass and has also been shown to prevent electrochemical corrosion in thin film modules [25]. Measurement with energy dispersive spectroscopy (EDS) showed that primary impurity in soda-lime glass is sodium with 16% by weight. As expected, borosilicate glass has a much lower sodium content at 6.5% by weight. PID was observed in both of one-cell coupons and confirmed by I-V and EL. Figure 3.9. shows that the borosilicate coupon is much more resistant against PID even at 100 h PID stress while the soda-lime glass coupon was completely shunted after 24 h PID stress. These results support that sodium in a glass plays an important role in causing PID. The borosilicate glass could be used in PV modules to minimize PID, but it cannot completely protect cells from PID and the cost would be significantly higher.



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Figure 3.9 PID (60 °C/85% RH, -600 V) Progress of Soda-Lime Glass Coupon and Borosilicate Glass Coupon

3.7 Summary

The PID-recovery methods presented in this chapter show very good regeneration (up to 96%) of P_{max} at a high irradiance level of 1000 W/m². However, the recovery of shunt resistance and efficiency at low irradiance levels is extremely low when compared with the recovery of P_{max} /efficiency at a high irradiance level. The poor cell efficiency at low irradiance levels would lead to lower energy production at predominantly low irradiance level locations and durations. The near-full P_{max} recovered cells still show much higher reverse-bias current when compared with the fresh cells. Therefore, any recovered cells/modules through the application of reverse potential and/or high temperature may pose safety risks under shaded condition if the protecting bypass diodes happen to fail in the field. Consequently, the P_{max} recovery at high irradiance level alone should not be considered as the sole parameter indicating the full recovery from the PID issue as the cells still retain some of the defects caused by the PID damage.

The two modern, commercial QE systems utilized in this work had substantial scaling error when measuring heavily shunted cells, such as PID-affected cells. The existing QE system with the low impedance accessory effectively minimizes (though not



totally eliminates) QE scaling error issues in the shunted cells. Consequently, it is highly recommended to use the QE systems built with the low impedance accessory, not only for the PID-related reliability research but also for the solar cell design research.

Two different glass types (soda-lime and borosilicate glass) were utilized to determine PID effects. It was shown that borosilicate glass has a lower sodium content (6.5%) than soda-lime glass (16%) by EDS analysis and that a borosilicate glass coupon is more PID resistant than using soda-lime glass. It is concluded that sodium in glass is critical in PID effect.



Chapter 4

QUANTUM EFFICIENCY LOSS CAUSED BY PID

4.1 Introduction

There are several studies indicating that the PID loss is caused by the migration of sodium ions from glass superstrate to the cell [19, 31, 43]. One of the primary solutions to address PID issue is a modification of chemical and physical properties of antireflection coating (ARC) on the cell surface [28, 37]. Depending on the edge isolation method used during cell processing, the ARC layer near the edges may be uniformly or non-uniformly damaged. Therefore, the pathway for sodium migration to the cell junction could be either through the cell surface if surface and edge ARC have low quality or through the cell edge if surface ARC is high quality but edge ARC is damaged [28]. In a previous study [92], a wavelength dependent QE (quantum efficiency) loss and recovery after PID stress testing was shown. In this study, two PID susceptible cells from two different manufacturers have been investigated. The QE measurements of these cells before and after PID stress were performed at both surface and edge locations to determine the influence of sodium migration pathway on the wavelength dependent QE loss. PID-affected QE curves based on QE loss at short wavelength presented in the previous chapter have also been modeled in this chapter. Investigation of the challenges with an alternate PID inversion model [40] is presented in this chapter. To rapidly screen a large number of cells for PID stress testing, a new but simple test setup that does not require laminated cell coupon has been developed and is used in this investigation.



4.2 Experiments

The solar cells used in this study were standard 156×156 mm p-base monocrystalline silicon cells with known susceptibility to PID. Two different manufacturers' cells (M1 and M2) were used in this investigation. Two test setups were used to perform PID stress testing. In the first setup, laminated one-cell coupons were used. In order to fabricate laminated one-cell coupons, each cell was encapsulated as in a standard PV module construction (glass – EVA – cell – EVA – backsheet) using a commercial PV module laminator. Typical tempered solar glass (soda-lime glass), EVA, and TPE backsheets were used in fabricating the laminated one-cell coupons. The PID stress conditions for the coupons were obtained using a weathering chamber at 60 °C/0% relative humidity (RH) or 85 °C/0% RH. Instead of maintaining commonly used 85% RH for the glass surface conductivity during PID test, an aluminum tape with conductive adhesive was used to cover the front glass surface of the coupons. The voltage of –600V was applied on the cell with respect to aluminum tape on a coupon glass. More detailed description of the setup is presented elsewhere [92].

In the second setup, simple physically stacked (no lamination) construction materials were used [31, 43, 100]. The laminated coupon based PID test setup is good to observe if a cell is PID susceptible or not; however, in this method several physical and surface analyses of the cell (before and after PID stress testing) using various characterization techniques would be limited due to difficulty in delaminating cell from other coupon materials especially encapsulant. Therefore, a simple and rapid PID test setup that does not require laminated coupons has been developed at Arizona State University. Same materials used for the laminated coupon method were utilized in this



new setup as well. The glass, EVA, and cell are stacked, as shown in Figure 4.1 left, and -600V was applied to the bottom plate. Using this new setup, the cells were PID stressed at 60 °C/0% RH in a conventional laboratory convection oven. By using this new method, we were able to secure, after the PID stress testing, the entire bare cell with no breakage or EVA residue on it.



Figure 4.1 Photo of Setup #2

All the laminated one-cell coupons and bare cells were characterized by light I–V, dark I–V, electroluminescence (EL) imaging, dark lock-in thermography (DLIT), and QE before and after PID tests. All QE values obtained after PID tests were normalized due to the scaling error [93]. QE measurements at multiple cell locations of interest were carried



out to investigate the wavelength dependent QE loss. Based on the lessons learned in the previous study [92], primary attention was paid to the QE loss observed at the short wavelength range (300 – 700nm). After PID stress and initial characterizations, all the degraded coupons and bare cells were stored at room temperature and monitored the recovery process as was done in the previous study [92].

4.3 Quantum Efficiency Loss Modeling

The unrecovered QE or QE loss could be attributed to change of front surface recombination velocity (FSRV) or emitter diffusion length [98, 101]. PC1D was used in simulating this effect, and QE loss shape of PID-stressed coupon in short wavelength was clearly demonstrated by either increasing FSRV or decreasing emitter diffusion length (not shown here). However, in a real PID cell, measuring area includes both PID-affected area and no/less PID-affected area due to a bigger beam size $(1 \text{ mm} \times 5 \text{ mm})$ of monochromatic light in QE system than shunting spots in PID-affected area. It has been observed that PID-stressed cell has localized shunting spots, which appear circularshaped with a diameter of 5-20 μ m [35]. Since PC1D simulates the homogeneous region only, it was considered that parallel connection of no/less PID-affected area assuming low FSRV and PID-affected area assuming high FSRV were considered as a measured area. Each area of QE is obtained by using the following equations [102], and parameters used in creating these two areas are shown in Table 4.1. To make the parallel connection of those two different areas, total QE (non-PID area + PID area) plot was calculated by multiplying a factor (for example, PID area: 0.7 and non-PID area: 0.3 were used).



$$QEE = \frac{\alpha L_p}{\alpha^2 L_p^2 - 1} \times \left[\frac{\frac{S_p L_p}{D_p} + \alpha L_p - e^{-\alpha x_j} \left(\frac{S_p L_p}{D_p} \cosh \frac{x_j}{L_p} + \sinh \frac{x_j}{L_p} \right)}{\frac{S_p L_p}{D_p} \sinh \frac{x_j}{L_p} + \cosh \frac{x_j}{L_p}} - \alpha L_p - e^{-\alpha x_j} \right]$$
(3.1)

$$QEB = e^{-\alpha x_j} \frac{\alpha L_n}{\alpha^2 L_n^2 - 1} \times \left[\alpha L_n - \frac{\frac{S_n L_n}{D_n} \left(\cosh \frac{H}{L_n} - e^{-\alpha H} \right) + \sinh \frac{H}{L_n} + \alpha L_n e^{-\alpha H}}{\frac{S_n L_n}{D_n} \sinh \frac{H}{L_n} + \cosh \frac{H}{L_n}} \right]$$
(3.2)

$$QE = QEE + QEB \tag{3.3}$$

where

- QEE: QE at emitter region
- QEB: QE at base region
- L_p: emitter diffusion length (µm)
- S_p: front surface recombination velocity (cm/s)
- D_p: emitter diffusivity (cm²/s)
- x_j : emitter thickness (junction depth) (μm)
- α : absorption coefficient (cm⁻¹)
- L_n : base diffusion length (µm)
- Sn: back surface recombination velocity (cm/s)
- D_n: base diffusivity (cm²/s)
- H: base thickness (µm)



	QE Parameters	QE Parameters
	for	for
	non-PID-affected area	PID-affected area
S _p (cm/s)	1×10^4	$5 imes 10^4$
L_{p} (μm)	1	1
$D_p (cm^2/s)$	4	4
x _j (µm)	0.5	0.5
S_n (cm/s)	1×10^4	$1 imes 10^4$
L_n (μm)	200	200
$D_n (cm^2/s)$	27	27
Η (μm)	200	200

Table 4.1Parameters for QE Plots

Figure 4.2 shows simulated QE results from the calculation mentioned above. Blue curve can be represented as QE of non-PID-affected homogeneous area. Calculated QE of PID-affected homogeneous area is shown as red curve, and it should be noticed that QE in short wavelength (300-700 nm) is decreased, as if QE of PID-stressed coupon shown in Figure 3.8 is plotted. QE of total area close to real measurement also shows such QE loss shape in the short wavelength. These results indicate that a cell property could be changed by defects caused by PID in addition to change of I–V parameters. Based on the results presented in Section 3.5, these defects leading to decrease QE in short wavelength may not be removed, although I–V parameters are getting slowly recovered. As shown in Figure 4.2, the modeled QE at 300-360 nm is clearer in discriminating QE shifting as FSRV changes. However, QE loss from one-cell coupon, as shown in Figure 3.8, has no values at such range due to cut-off wavelength (360 nm) of



EVA. Therefore, it is recommended to measure QE of PID-affected area without EVA and glass to verify this QE model.



Figure 4.2 Calculated QE to Simulate PID-Affected Cell. Blue Curve is based on Homogeneous Non-PID-Affected Area, Red Curve is based on Homogeneous PID-Affected Area, and Green Curve is based on a Combination of 70% PID-Affected Area and 30% Non-PID-Affected Area.

4.4 Detailed Modeling of Surface Charge with Sentaurus

The previous section discussed the effect of changing surface recombination on QE. It has also been postulated that the sodium ions cause shunting though surface charge. According to proposed mechanism [40] the large potential difference between the cell and the frame causes a large amount of sodium to move from the glass, through the EVA encapsulant and accumulate at the surface of the solar cell. The huge amount of positively charged sodium ions at the cell surface attract the same amount of negative charges within the silicon to invert the emitter and eventually shunt the cell [39, 40]. The



researchers estimated the amount of surface charges required to show the emitter inversion for typical n⁺ emitter with a sheet resistance of ~ 60 Ω/sq is -1×10^{15} cm⁻², but this amount of charges was not able to be applied in PC1D [40]. In this study, a different semiconductor simulation program, Sentaurus allowing for much higher surface charge to investigate the proposed mechanism, was applied. In the modelling, typical industrial solar cell parameters were used, with a p-type base doping of 1×10^{16} cm⁻³, a peak ntype emitter doping of 1.5×10^{20} cm⁻³ and a junction depth of 0.5 µm. Figure 4.3 shows the band diagram when surface charge of -1×10^{15} cm⁻² was applied on an emitter surface. There was no complete emitter inversion that was shown in [40]. The beginning of emitter surface started to invert, however the inversion does not extend down to the junction region as would be required for the inversion to cause the cell to shunt. Increasing the charge further produced a similar effect to that shown in Figure 4.3. The surface inverts and would change the surface recombination velocity but the effect extends less than 10 nm into the emitter and is nowhere near the junction at $0.5 \,\mu$ m. Even with an unrealistically high level of -1×10^{30} cm⁻² charge applied on the emitter surface the cell does not exhibit shunting. The Sentaurus simulation confirms the results from Saint-Cast et al., who reported, by comparison of experimental and simulated dark I-V curves, that the inversion model cannot explain the PID shunting mechanism [99]. Therefore, a recently proposed model based on sodium decorated stacking faults [35] is preferred to the inversion model to explain the PID mechanism.





Figure 4.3 Sentaurus Band Diagram for a Solar Cell, (a): No Surface Charge, (b): with Negative Surface Charge $(-1 \times 10^{15} \text{ cm}^{-2})$

4.5 QE Loss Observed on Laminated Coupons (Setup #1)

As shown in Figure 4.4 (a), the maximum power (P_{max}) and shunt resistance (R_{sh}) values were found to be dramatically decreased after PID stress at 85 °C/–600 V/44 h. P_{max} was recovered to approximately 90% of initial while R_{sh} was recovered at an extremely slow rate. A significant QE loss was observed, at different locations on the cell



surfaces including the cell edges, at the wavelength range between 360 and 700 nm. It is important to note that the QE loss seen at this low wavelength range was never recovered while P_{max} and R_{sh} were slowly and partially recovered with respect to time. A separate mechanism might therefore be responsible for the blue response loss. In order to determine if the QE loss in the low wavelength region is caused by the reflection alterations (reflection or absorption) in the ARC layer, the UV-Vis-NIR reflectance spectra were obtained on a different coupon stressed at 60 °C/–600 V/88 h. Negligible reflectance spectral difference was observed between pre- and post-PID stress test, as shown in Figure 4.5. The QE loss in the short wavelength range could be attributed to increasing front surface recombination velocity (FSRV) or decreasing emitter diffusion length [101], and it was modelled in Section 4.3.





Figure 4.4 I–V Parameters (a) and QE (b) Results of a Laminated One-Cell (M1; Setup 1) Coupon at Various Stages of Investigation: PID 44 h at 85 $^{\circ}$ C/0% RH, -600 V, Recovery at Room Temperature (RT).





Figure 4.5 Reflectance of a Laminated One-Cell (M1; setup 1) Coupon at Various Stages of Investigation: PID 88 h at 60 °C/0% RH, -600 V, Recovery at Room Temperature (RT). Zoomed-In Plot of Reflectance is shown in a Small Square in Figure 4.5.

4.6 QE Loss Observed on Bare Cells (Setup #2)

In the FSRV model, the QE change at 300 nm is expected to be more sensitive as compared to the higher wavelength region as presented in Section 4.3. Therefore, if we observe a higher loss at about 300 nm in the PID-affected cell, it would give us additional confidence if the FSRV or decreasing emitter diffusion length model is operating after PID stressing. Measuring QE at 300 nm on the laminated one-cell coupon was not possible due to UV cut-off wavelength (360 nm) of EVA. Therefore, a new PID test setup (setup #2) without lamination that would allow QE characterization up to 300 nm after PID was needed.

Figure 4.6 shows QE results of a PID stressed cell (M1) obtained using the new method. It is clear that QE loss at 300nm in largest, and this experimental result is consistent with the one of QE model, as shown in Figure 4.2. Higher QE loss at about



300 nm appears to indicate that the FSRV mechanism is probably responsible for the PID losses.



Figure 4.6 QE of Bare Cell (M1; setup 2) before and after PID 80 °C/24 h/–600 V. QE of PID Affected Area Characterized by EL was carried out. QE of PID Stressed Cell is normalized.

As shown in Figure 4.7, PID was observed almost exclusively at the edges only for M2 cells irrespective of laminated coupon method (setup 1) or bare cell method (setup 2) is used. PID was observed all over the surface of M1 cells but, on M2 cells, it was observed essentially at cell edges only as shown in the exploded EL images of Figure 4.8 (a). It is believed that M2 cells have appropriate ARC layers to prevent PID but presumably get damaged at the edges during edge isolation process which is commonly done after diffusion in cell processing lines or the cell had an inadequate edge isolation process [28]. QE measurements, before and after PID stress, was carried out on areas where PID was observed (Area 1) and also on areas where no PID was observed (Area 2)



based on the EL images. Neither area had wavelength dependent QE loss after normalization (as opposed to M1 cells), as shown in Figure 4.8 (b). These results from M2 cell appear to indicate that there is another pathway operating for sodium ions to reach the cell junction without going through ARC layers. If the edge isolation process is inadequate or inappropriately damaged, junction near the edges is susceptible to PID loss but the loss would be wavelength independent as the photons do not go through ARC layer and the FSRV mechanism is not operating.





Figure 4.7 Comparison PID Test between (a) Laminated Coupon Method (Setup 1) vs. (b) Bare Cell Method (Setup 2). M2 Cells were used in Both Methods.






(a)



(b)

Figure 4.8 EL Image before/after PID and QE in 2 Areas. M2 Cell was used.

4.7 Summary

QE loss in short wavelength was simulated by increasing FSRV, and it showed a similar QE shape as the QE of actual PID-stressed coupons. Result of bare cell QE measurements enabling QE at around 300 nm clearly matches with the calculated QE loss model. Depending on edge isolation adequacy or damages of ARC during edge isolation at the edges, cells may experience PID near edges even if the front surface is well



protected against PID by appropriately modifying the ARC layers. A simple and rapid PID test method is used, which allowed us to carry out the PID experiments with no requirement of laminated cell coupon.



Chapter 5

EFFECT OF SIN_X REFRACTIVE INDEX AND EMITTER SHEET RESISTANCE ON POTENTIAL-INDUCED DEGRADATION

5.1 Introduction

Potential-induced degradation (PID) has become a hot issue in the photovoltaic (PV) reliability industry due to the likelihood of rapid power degradation. The term PID was first used by Solon in 2005 [28]. Since its main failure mechanism manifests itself in cell shunting, the PID is specifically referred to as PID of the shunting type (PID-s) [35, 43]. PID-s is only observed on p-type diffused junction solar cells when the cells are operated as a PV system which is accompanied by high voltage with severe environmental conditions, such as high temperature and humidity. Several studies indicate that PID-s is caused by the migration of sodium ions, and the source of sodium could be from the PV module glass (typically, soda-lime glass) [19, 33, 44, 103] or contaminants at the cell surface [104]. A more recent study showed that a root cause of PID-s is the presence of a sodium-decorated stacking fault penetrating the n-p junction [35]. The sodium-decorated stacking faults cause junction shunting, and this leads to an increase of the depletion region recombination current (J_{02}) and the ideality factor (n_2) [43]. It also has been reported that sodium outdiffusion from the stacking fault is observed as the PID-affected cell is recovered by exposure to high temperature [48].

There are multiple factors that contribute to PID susceptibility in a cell process. What we would like to clarify among them are the roles of the refractive index (RI) of the silicon nitride (SiN_x) antireflection coating (ARC) and the emitter sheet resistance. Many



publications have shown that PID is suppressed by increasing the RI to a value greater than 2.1 [28, 36, 37]. Pingel *et al.*, showed the general trend of increasing PID-susceptibility with respect to increasing emitter sheet resistance [28]. In this study, we verified the effect of SiN_x RI and emitter sheet resistance on PID-s.

5.2 Experiments

Industrial standard (156 mm × 156 mm) p-type diffused junction monocrystalline silicon solar cells were fabricated in the Solar Power Laboratory (SPL) pilot line located at Arizona State University. The SPL standard p-type cell process flow is shown schematically in Figure 5.1. For this study, three different SiN_x ARC films were deposited onto wafers by adjusting the SiH₄ and NH₃ gas flow rates in the plasma enhanced chemical vapor deposition (PECVD) equipment. All of the ARC film thicknesses are around 78 nm. The cells were processed identically except for the change in refractive index shown in Table 5.1. Variable angle spectroscopic ellipsometry (VASE) characterized the RI and film thickness of the ARC films deposited onto 6-inch round single-side polished wafers. A corona charging method followed by time-resolved surface voltage measurement [105] was used to determine the PID-s susceptibility of SiN_x ARC films deposited on finished solar cells and polished wafers. Ion drift (ID) spectrometry [45] was utilized to verify the sodium contamination of test cells before PID testing.



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Figure 5.1 SPL Standard Cell Process Flow

Sample Group	SiN _x ARC RI	Emitter Sheet Resistance (Ω/sq)	Bubbler Temperature (°C)
А	1.87	~60	30
В	1.94	~60	30
С	2.05	~60	30
D	1.94	~60	30
Е	1.94	~70	30
F	1.94	~70	20
G	1.94	~80	30
Н	1.94	~80	20

For the study regarding the impact of emitter sheet resistance on PID-s, various emitter sample groups were prepared (shown as D-H in Table 5.1). These included the SPL standard diffused emitter with a sheet resistance of ~60 Ω /sq (Group D), and four emitters with higher sheet resistance (Group E-H) than the SPL standard emitter. The



higher sheet resistance emitters were formed by decreasing the SPL standard diffusion temperature from 840 °C to 800 °C and lowering the bubbler temperature. A four-point probe measured the emitter sheet resistance after removal of phosphosilicate glass (PSG) created during the diffusion process. The phosphorus concentration in the SPL standard emitter was profiled by secondary ion mass spectrometry (SIMS). Electrochemical capacitance-voltage (ECV) was also carried out on sample groups F and H to compare with the SPL standard emitter.

One-cell coupons laminated by using typical commercial PV module materials (soda-lime glass, EVA, TPE backsheet) were prepared to carry out some of the PID stress experiments. These coupons were PID stressed under conditions of 60 °C/85% relative humidity (RH) and -600 V in an Atlas Ci4000 weathering chamber. An alternate PID test setup that does not require coupon lamination was also used in this study to prepare samples for analysis (e.g., SIMS measurements) after stress testing. The cell and PV module materials are simply stacked up for PID stressing [31, 43, 100], so the PID-stressed cell is easily removed from the module materials without the need to employ a complicated delamination process. Figure 5.2 graphically depicts the alternate test setup used in this study. We have previously shown that conventional laminated cells and ones unlaminated such as used in the present study result in a similar grade of PID when stressed [106]. PID stress was carried out for these cells under 60 °C/0% RH and -600 V in a scientific oven.





Figure 5.2 PID Test Setup without Module Lamination

Light current-voltage (I–V) and dark I–V were measured before and after the PID tests. A steady state solar simulator was used for the I–V measurements, and the light I–V was taken at both standard test condition (STC), which is 1000 W/m² irradiance and 25 °C cell temperature, and at low irradiance (LI, ~250 W/m²) conditions. Electroluminescence (EL) image was acquired via a coolSamBa HR-830 Si camera with concomitant imaging software. Quantum efficiency (QE) was performed on a QEX10 tool (PV Measurements, Inc).

5.3 Effect of SiN_x Refractive Index

Pervasive in the literature, PID has been observed in cells that have a low RI (<2.1) SiN_x ARC [28, 36, 37]. In this study, the three samples shown as groups A-C in Table I, including an SPL standard cell, all have lower refractive indices than 2.1. Thus, it was expected to observe PID on all the cells after PID stressing (60 °C/85% RH, -600 V, 96 h). Figure 5.3 shows the PID results of those three samples. Contrary to the work cited above, none of the samples showed any maximum power point (P_{max}) degradation after PID 96 h, regardless of the RI. Even the cell that with a very low RI (Group A, RI: 1.87) showed no P_{max} decrease at all. The only change seen was a decrease in the shunt





Figure 5.3 P_{max} and R_{sh} Progression in PID test (60 °C/85% RH, -600 V, 96 h) Cells with Different Refractive Indices. They were encapsulated as One-Cell Coupons with a PV Laminator (a) Group A: RI 1.87 SiN_x ARC, (b) Group B: RI 1.94 SiN_x ARC, (c) Group C: RI 2.05 SiN_x ARC.

resistance (R_{sh}) of group A to 60% of its initial value, as shown in Figure 5.3 (a). The SPL standard sample (Group B) that has a RI of 1.94 showed a lower decrease of R_{sh} after PID stressing than what was observed for the Group A cell with RI 1.87. The



highest RI cell (Group C) showed no degradation in either P_{max} or R_{sh} , as shown in Figure 5.3 (c). Note that the initial R_{sh} for this cell is already very low as compared to Group A and B cells. This is attributed to non-optimization of the firing conditions for the RI 2.05 SiN_x film. That is to say that we only varied the RI from our standard cell (Group B) process and left all other processing steps, including firing, the same in order to minimize experimental variables, which might cause other effects on PID-s. The low initial R_{sh} likely could be fixed by optimizing the firing process for each SiN_x film. In terms of R_{sh} , this result is consistent with prior experimental observations showing stronger PIDresistant cells with increased RI [28, 36, 37]. In order to verify the PID-susceptibility of those SiN_x ARC films, a method employing high dose corona charging followed by timeresolved measurement of surface voltage [105] was carried out for both polished wafer samples and solar cell samples without lamination. The result is presented in Figure 5.4. Wilson et al., showed that PID-susceptibility correlates with a higher surface voltage, which is dependent on SiN_x RI, after corona charging, and their RI 1.9 SiN_x ARC cell experienced 68% module efficiency decrease due to the PID [105]. As shown in Figure 5.4, the surface voltage after corona charging increases as RI decreases from both samples, and this trend is consistent with Wilson et al.'s result. Besides, the retained surface voltage after corona charging of RI 1.87 and 1.94 SiN_x films on final solar cell is very high (~ 30 V) as compared to that of the RI 2.05 SiN_x film. It means that our RI 1.87 and 1.94 SiN_{x} films are PID susceptible enough. But, another factor prevents the progress of PID-s, which causes a very slow R_{sh} decrease. It shows that the cells fabricated in the SPL pilot line are extremely PID-resistant regardless of the RI of the SiN_x ARC film due to another factor. The reason could be attributable to a low emitter sheet resistance, which



is about 60 Ω /sq. Pingel *et al.*, reported that the emitter sheet resistance is one of the factors which influences PID susceptibility [28] and that PID-susceptibility is decreased as emitter sheet resistance decreased. Therefore, a further study regarding the effect of emitter sheet resistance on PID-s was carried out, which is presented in Section 5.4. We also carried out non-contact ion drift (ID) spectrometry to measure mobile ions and determine the source of sodium ions causing PID-s [45]. One of the fresh (non-PID stressed) group A cells that was not encapsulated was chosen for this measurement. Initially, no mobile ions, especially sodium ions, were observed from this sample since only a small monotonic decrease of the surface voltage with increasing temperature was observed, as shown in Figure 5.5. After contamination with sodium ions, there was a reduction of the surface voltage with increasing temperature. The transition at 110 °C and a peak temperature (T) of 150 °C in dV/dT versus T spectra (not shown here) confirms the presence of sodium ions in the nitride [45]. The measurements indicate that cell has no sodium contamination as a consequence of the cell manufacturing process and support the hypothesis that the source of the sodium is the module (soda-lime) glass.





Figure 5.4 Positive and Negative Surface Voltage after Corona Charging on SiN_x Films with Various Refractive Index. Different Color Bars in a Group of RI Represent Different Times after Corona Charging Cessation. (a) SiN_x Film on Polished Wafer, (b) SiN_x film on Finished Solar Cell.



Figure 5.5 Sodium Ion Drift Characteristics of SiN_x Film (RI: 1.87) Used in This Study for Initial Surface and Intentional Sodium Contamination Surface.



5.4 Effect of Emitter Sheet Resistance

The cells in the previous section did not show PID, even after stressing for 96 hours, despite being fabricated using SiN_x ARC layers that would be expected to show PID degradation. As was mentioned previously, the SPL standard cell has an emitter sheet resistance of around 60 Ω /sq and higher emitter sheet resistance has been shown to increase PID susceptibility [28]. Therefore, cells having higher than 60 Ω /sq emitters were fabricated by modifying the POCl₃ diffusion process to incorporate lower diffusion temperatures and bubbler temperatures, as shown in Table 5.1. A total of five sample groups were prepared with various emitter sheet resistances (~60 Ω/sq , ~70 Ω/sq , ~80 Ω /sq). The SPL standard SiN_x ARC (RI: 1.94) was deposited onto these samples. PID test samples were prepared without lamination in case further cell analysis was warranted. As shown in Figure 5.2, the cells were stacked with commercial PV module materials and two metal plates. The stacked setup was placed in an oven to maintain a temperature of 60 °C during PID stressing. The cells with 60 Ω/sq (Group D) did not show PID whereas PID-s was clearly observed in cells with possessing higher emitter sheet resistance > 60 Ω /sq (Group E-H). Figure 5.6 shows EL images of a group H cell and a group F cell before and after PID stressing. Interestingly, the group H cell has a darker shunted area than the group F cell in the EL images, although both cells showed nearly the same degree of degradation, which is $\sim 40\%$ P_{max} decrease and $\sim 80\%$ R_{sh} decrease (not shown here). In addition, it was observed that the pattern of the shunted region as seen on the EL images in Figure 5.6 (b) and (d) is very similar. This phenomenon might be attributed to a difference of sheet resistance across the cell caused by diffusion process. A total of sixteen points were measured by four-point probe after the diffusion process, and this was



followed by the remainder of the standard SPL cell fabrication process. Overall emitter sheet resistance of one of the Group G cell was ~80 \pm 5 Ω /sq. Such a cell was stressed at 60 °C/0% RH, -600 V, 23 h, and then EL imaging was performed. Each point at which sheet resistance was measured was carefully investigated with EL imaging after PID; however, there was no clear correlation between PID-susceptibility and sheet resistance within a cell. A more detailed study with high resolution mapping of the sheet resistance using an automated tool might help to determine the relationship.



Figure 5.6 EL Images of Different Emitter Sheet Resistance Cells from SPL Pilot Line before/after PID Stress (60 °C/0% RH, -600 V, 23 h). (a) before PID of R_{sheet}: 80 Ω /sq Cell (Group H), (b) after PID of R_{sheet}: 80 Ω /sq Cell (Group H), (c) before PID of R_{sheet}: 70 Ω /sq Cell (Group F), (d) after PID of R_{sheet}: 70 Ω /sq Cell (Group F)

The uniformity might affect the PID-s susceptibility in addition to the value of the sheet resistance [28], so it was necessary to look into the sheet resistance uniformity of cells from all of the different diffusion processes. The emitter sheet resistance was measured on all the wafers from different groups of each batch (~ 13 wafers), as shown



in Figure 5.7. The sheet resistance distribution of SPL standard emitter (Group D) that has shown no PID-s is shown in Figure 5.7 (a). It has quite a good uniformity within the wafer and across the boat. The sheet resistance range within the wafer is about $\pm 2 \Omega/sq$. The other diffusion processes that were used in making higher emitter sheet resistance than SPL standard emitter result in a high non-uniformity of sheet resistance within the wafer, as shown in Figure 5.7 (b) and (c). But, cell groups with larger non-uniformity than group D did not always exhibit strong PID-s. Group E showed very weak PID-s (see Figure 5.8) although it has similar range of emitter sheet resistance with group F, as shown in Figure 5.7 (b). Therefore, the emitter sheet resistance uniformity cannot be the only criteria determining PID-s susceptibility.









Figure 5.7 Emitter Sheet Resistance Distribution. (a) SPL Standard Diffusion Recipe, (b) Sample Group E and F, (c) Sample Group G.



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Figure 5.8 PID-Susceptibility with Different Emitter Cells. Values are Post-PID Stress (60 °C, 23 h).

Another interesting result is that PID-s was also affected by the bubbler temperature in the emitter diffusion process. Regardless of the bubbler temperature, both group E and F showed a sheet resistance of ~70 Ω /sq. However, the cells from group E showed very weak PID-s while the group F cells showed strong PID-s. By increasing the bubbler temperature, the depth of the electrically active phosphorus plateau as determined by electrochemical capacitance-voltage (ECV) is increased [107]. It is speculated that PID-s is suppressed by increasing the electrically active phosphorus plateau near the surface in the emitter. A pair of group E and F and a pair of group G and H cells have similar emitter sheet resistance ranges, respectively. However, group F showed stronger PID than group E, and group H showed higher PID degradation than group G, as shown in Figure 5.8. This result clearly shows an effect of bubbler temperature on PID susceptibility as well as an effect of emitter sheet resistance. In addition, the SPL standard cell (Group D) has extremely high total surface phosphorus concentration (6 ×



 10^{21} cm⁻³) as well as a higher and deeper electrically active phosphorous plateau than Group F and H cells, as shown in Figure 5.9. Such a high phosphorus concentration layer could work as a sodium gettering layer [108]. Higher than 1×10^{20} cm⁻³ of dopant surface concentration enhances the effectivity of impurity gettering [109]; for example, platinum gettering has been observed where phosphorus $(1 \times 10^{21} \text{ cm}^{-3})$ was diffused [110]. To verify the effect of phosphorus concentration on sodium gettering, solar cell samples using Group D (PID-resistant) and F (PID-susceptible) conditions were prepared for SIMS analysis. 6-inch round single-side polished wafers were used as substrates in fabricating these solar cells for the SIMS measurement. Texturing and front metallization was omitted for simplicity, but the SiN_x ARC and back surface field (BSF) were processed for PID-s characterization since they are necessary in observing PID-s [111]. The cell was cut into several pieces to divide into PID testing samples and non-PID testing samples. Subsequently, PID stressing was applied to those PID testing samples using the stacked method shown in Figure 5.2. Figure 5.10 shows the SIMS results of sodium concentration in the cells. In this SIMS, a concentration value lower than 1×10^{16} cm⁻³ is regarded as background noise. It should be noticed that the sodium has diffused more deeply in the PID-susceptible sample (Group F) with a phosphorous concentration of 4×10^{20} cm⁻³ than in the PID resistant sample (Group D) with a phosphorous concentration 1.4×10^{21} cm⁻³ after PID stressing, as shown in Figure 5.10. Note that the junction depth for these samples is approximately 0.4 µm. For the PID-resistant cell with a high phosphorus concentration emitter (Group D), more of the sodium ions remain in the SiN_x layer and the top of the emitter layer as shown in Figure 5.10 (b). This result shows that sodium transport could be slowed down or blocked by a layer with a high



phosphorous concentration. It also has been reported that PID-s was suppressed by keeping the PSG layer (which has high phosphorus concentration) after POCl₃ diffusion in a cell process [66]. Therefore, PID-s could be suppressed by the presence of a high surface phosphorus concentration region/layer in the emitter, which might work as a sodium gettering layer. Moreover, Figure 5.10 shows that the non-PID stressed sample has a very low sodium concentration while the PID-stressed sample has higher than 1×10^{22} cm⁻³ of surface sodium concentration. This is further evidence that the source of sodium causing PID-s could be solar (soda-lime) glass.



Figure 5.9 ECV and SIMS Profiles for Different Solar Cell Emitters.





Figure 5.10 SIMS Profiles for Cells Fabricated from Group F and D. PID Stressing at 60 °C/0% RH, -600 V, 96 h (a) Solar Cell with $\sim 70 \Omega$ /sq Emitter Sheet Resistance. (b) Solar Cell with $\sim 60 \Omega$ /sq Emitter Sheet Resistance.

5.5 Summary

P-type cells fabricated in SPL at ASU with various SiN_x RI ARCs were PID stressed for 96 hours. As confirmed by ID spectrometry and SIMS, no sodium contamination was introduced by the fabrication process before PID testing. None of those cells showed P_{max} degradation and strong shunting, as confirmed by I-V and EL. But, a slight decrease of shunt resistance that does not affect P_{max} was observed from low



RI cells (RI: 1.87 and 1.94). The SiN_x film of these low RI cells was determined to be PID-susceptible, but the cells are extremely PID-resistant regardless of RI. One possible explanation could be due to the extremely high surface phosphorus concentration in the emitter (60 Ω/sq) of SPL-fabricated cells. The cells that have higher emitter sheet resistance (70 Ω /sq and 80 Ω /sq) were fabricated by modifying diffusion temperature and were subsequently PID-stressed. Some of these cells showed strong PID-s while the 60 Ω /sq cell had no PID-s. This result is consistent with Pingel *et al.*'s result [28]. However, cells with a diffusion process resulting in $\sim 70 \Omega/sq$ emitters did not always demonstrate PID-s. On one hand, a 70 Ω /sq emitter cell fabricated using a 30 °C bubbler temperature in a diffusion process showed extremely weak PID-s while one fabricated using a 20 °C bubbler temperature showed strong PID-s. Change of the bubbler temperature in the POCl₃ diffusion process results in a change of the electrically active phosphorus surface plateau depth, which could affect PID susceptibility. SIMS results for a PID-resistant cell showing high sodium concentration near the emitter surface after PID stressing support the theory of sodium gettering by emitters with a high phosphorous concentration. It was shown that emitter sheet resistance alone could not determine the PID-susceptibility. Therefore, it is suggested that the active/inactive phosphorus concentrations as well as the emitter sheet resistance should be both carefully monitored and controlled when manufacturing p-type crystalline silicon solar cells in order to avoid PID-s.



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CHAPTER 6

A NOVEL TECHNIQUE FOR PERFORMING PID SUSCEPTIBILITY SCREENING DURING THE SOLAR CELL FABRICATION PROCESS

6.1 Introduction

One issue in the area of photovoltaic (PV) reliability that has garnered significant scrutiny over the last several years has been potential induced degradation (PID). It has been observed that the propensity for a cell to be susceptible to PID can result in a rapid output power degradation in a short period of time [28]. The PID is caused by a negative potential to the cell with respect to ground under humid environmental condition. The junction shunting of p-type crystalline silicon solar cells has been recognized as a failure mechanism, so this PID is also referred to as the shunting type of PID (PID-s). Recent literature revealed both the presence of sodium in the silicon nitride (SiN_x) anti-reflection coating (ARC) layer and emitter, as well as stacking faults across the n-p junction in PIDstressed cells/modules [35, 43]. Several PID mechanisms have been suggested, and currently, the most probable PID mechanism is that the sodium-decorated stacking faults across the junction cause the junction shunting, which is PID-s [35, 43, 48]. The source of the sodium could be either from soda-lime glass that is commonly used as a PV module glass [19, 33, 103] or the contaminants entrained during the cell fabrication process [104].

Since its failure mechanism is shunting, various characterization techniques can be utilized to detect PID-s. The shunt resistance is conveniently obtained by light or dark current-voltage (I–V) measurements. An I–V curve also visually depicts a change of



shunt resistance by examining the slope near the short-circuit current (I_{sc}) point. Electroluminescence (EL) imaging portrays areas of PID shunting by representing them with reduced brightness. Dark lock-in thermography (DLIT) is also used to detect PID shunted areas, which experience higher temperature than non-shunted areas. Finally, quantum efficiency (QE) might be used to visualize PID QE loss in addition to PID shunting [103]. All of the aforementioned PID-s characterization techniques require solar cells with complete contacts, that is a finished cell having the metal contacts, in order to detect PID-s degradation. Notwithstanding this, it would be favorable to screen for PIDsusceptibility during the cell fabrication process as a means of reducing cost and saving time. Accordingly, we utilized illuminated lock-in thermography (ILIT) which allows for shunt detection on solar cells that have no contacts [112]. An advantage of using ILIT during cell processing is not only screening for PID-s but also visualizing shunting that is a consequence of the particular cell processing employed [112, 113]. In this paper, we present a technique that does not require the metallization in evaluating the PIDsusceptibility.

6.2 Experiments

Standard $156 \times 156 \text{ cm}^2$ p-type monocrystalline silicon solar cells were fabricated in the Solar Power Laboratory (SPL) at Arizona State University. The p-type wafers were textured with a KOH solution and followed by a wafer cleaning process. Backside oxide deposition was carried out by plasma enhanced chemical vapor deposition (PECVD) in order to prevent the formation of an emitter on the backside of the wafers. POCl₃ diffusion was performed in an MRL furnace followed by a buffered oxide etch (BOE) which simultaneously removed the phosphosilicate glass (PSG) and the backside oxide.



The SPL standard SiN_x ARC that has a refractive index of 1.94 and a thickness of 76 nm was deposited on the emitter by PECVD. When required for particular experiments, cell metallization contacts were formed by screen printing, which includes an aluminum back surface field (BSF) and front silver fingers and busbars.

PID testing was carried out using the stacked method, which does not require cell lamination [106]. Commercial PV module materials (tempered soda-lime glass and EVA) were appropriately stacked with a single cell (top metal plate – glass – EVA – cell – bottom metal plate). A voltage of –600V was applied to the bottom metal plate contacting a cell to simulate PID conditions. The cells were PID-stressed at 60 °C/0% relative humidity (RH) in a scientific oven for periods of time up to 96 hours.

Various characterization techniques were used in this study to determine if PID was present after stressing. A steady state solar simulator was used to take the I–V measurements. EL images were taken at forward bias of the cell (8 A current limit). Photoluminescence (PL) imaging was also carried out. DLIT images were obtained with forward (+0.5 V) and reverse (-0.5 V) biases. For ILIT measurements, the cells were light biased with an LED with output at 850 nm. All characterization was carried out before and after PID stressing.

6.3 PID-s Detection using Illuminated Lock-in Thermography

Figure 6.1 shows EL and LIT images of the cell after exposure to a 23-h PID stress at 60 °C/0% RH. The dark areas in the EL image shown in Figure 6.1 (b) represent the shunting generated by PID stressing. This shunting was also confirmed by I–V and LIT. As shown in Figure 6.1 (f), the red I–V curve clearly indicates shunting behavior of the stressed cell. By comparing results at a positive bias with those at the negative bias,



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DLIT images (Figure 6.1 (c) and (d)) show that the PID-s observed in the cell used in this study exhibits linear (ohmic) shunting since the shunt shows same signal (brightness) under forward and reverse bias [112]. Figure 6.1 (e) shows an ILIT image which was taken without voltage bias or contacting during the ILIT measurement process. It should be noticed that the signature of the PID-affected area is identical regardless of images from EL, DLIT, and ILIT. The ILIT image in Figure 6.1 proves that PID-s can be characterized without contact. Given this finding, the ILIT technique has the potential to be very useful in screening for PID-susceptible cells on substrates that have been processed up to, but not including screen printing. These results will be presented in the following section.





Figure 6.1 Images before and after PID Stressing at 60 °C/0% RH, -600 V, 23 h. (a) EL Image before PID Stressing, (b) EL Image after PID Stressing, (c) DLIT Image under +0.5 V, (d) DLIT Image under -0.5 V (e) ILIT Image (f) I–V curve

6.4 PID-s Screening during Cell Process

In order to determine the simplest sample structure that can be used to show PID-s after stressing, four different samples were prepared for this study. As mentioned in Section 6.2, the SPL standard cell process was carried out up to the diffusion process, and the wafers were subsequently separated into four groups, as shown in Table 6.1. Since the ILIT technique proved to viable for imaging PID-s without making contact during the measurement, cells were prepared without metallization. None of the cells has front



fingers/busbars, and the aluminum BSF was even omitted in groups C and D to simplify the test sample structure. Also the cells without a SiN_x ARC (groups B and D) were fabricated to determine if that film is necessary for showing PID-s utilizing the test method mentioned in the previous section. And then, PID stressing (60 °C/0% RH, -600 V, ~96 h) was applied to the cells from each group. For this study, ILIT was the only PID-s characterization technique employed before and after stressing since there is no way to contact cells for EL or I–V measurements for samples without either front or back metallization. Figure 6.2 shows the PID test results from the group A cell. After PID stressing, the ILIT image obviously showed PID shunting, as shown in Figure 6.2 (b). In this case, the shunted area was generated around the edge of the group A cell, while the cell having full metallization (Figure 6.1) has a localized shunted area. It is speculated that the omission of the fingers and busbars causes unequal potential above the cell surface under PID test conditions when compared to the potential above a cell having front contacts. This situation may cause different PID shunting pattern. PL, which does not need contacting during the measurement like ILIT measurement, was also used to image the PID-s, however the PL image could not clearly discriminate the shunting, as shown in Figure 6.2 (c) and (d). Therefore, ILIT is more suitable technique in screening PID-s for cells having not front metallization, such as group A cells. For further test sample simplification, we prepared and tested the cell without a SiN_x ARC (group B). No PID-s was observed in this cell, and this result is consistent with prior experimental observations showing no PID on cells without the SiN_x ARC [34]. Additionally, none of the cells from Group C and D that do not have an aluminum BSF experienced the PID-s regardless of the presence of a SiN_x ARC layer. Therefore, it is recommended that the



cell should have the SiN_x ARC and the aluminum BSF in order to see PID-s degradation in the PID test setup.

	SiN _x ARC	Fingers/busbars (Front Silver)	Aluminum BSF
Group A	Yes	No	Yes
Group B	No	No	Yes
Group C	Yes	No	No
Group D	No	No	No

Table 6.1Test Sample Groups Used in This Study



Figure 6.2 ILIT and PL Images of the Group A Cell, (a) ILIT before, (b) ILIT after (c) PL after (d) PL (color scale) after PID Stressing at 60 °C/0% RH, -600 V, 96 h.



One interesting thing we observed in this study is that there was no PID-s in areas where a four-point probe (4PP) measurement was carried out to determine emitter sheet resistance. The 4PP measurements were done after the post-diffusion BOE glass strip. Referring to Figure 6.3, the enlarged image of the cell used in Figure 6.1 and Figure 6.2 clearly shows no PID-affected areas where sheet resistance was measured. It is speculated that some impurities, for example, oxide, were generated when the probes of the 4PP tool touched the cell. It has been reported that an oxide layer between the SiN ARC and the emitter prevents PID [63, 65]. So, an oxide layer generated by the 4PP might protect the emitter from PID shunting. Further study regarding this observation is in progress.



Figure 6.3 Circles in the Images Indicate No PID-Affected Area where 4PP Measured. (a) Enlarged EL Image Shown in Figure 6.1 (b), (b) Enlarged ILIT Images Shown in Figure 6.2 (b), Scale was adjusted to Clearly Show the Effect of 4PP Measurements.

6.5 Summary

A PID-s screening method based around using ILIT was presented. The ILIT is a convenient technique showing PID shunting on cells without contacts. Using this advantage, PID-susceptible cells could be screened by the stacked PID test method and



ILIT at an earlier stage using cells having no front metallization. SiN_x ARC and aluminum BSF are necessary in order to observe the PID shunting.



CHAPTER 7

SURFACE DISRUPTION METHOD WITH FLEXIBLE GLASS TO PREVENT POTENTIAL-INDUCED DEGRADATION IN PV MODULES

7.1 Introduction

Potential-induced degradation (PID) is a performance degradation caused by a high negative voltage difference between the photovoltaic (PV) cells and the aluminum frame under humid environmental conditions. PID has a serious impact, on the PV system durability over a short period of time [28, 114]. PID caused by junction shunting is specifically referred to as the PID of the shunting type (PID-s) [35, 43]. Although the mechanism responsible for PID-s is not yet fully understood, many scientific studies have suggested that the migration of sodium ions from the PV module glass, such as soda-lime glass, to the cell causes PID-s [19, 33, 36]. Various methods to prevent or minimize PIDs have been developed and applied at the cell, module, and system levels. These include modification of the anti-reflection coating (ARC) on the cell surface to prevent PID-s at the cell level [28, 36, 37]. At the system level, modules affected by PID-s during daytime can be recovered by applying the opposite potential at night [36]; however, 100% recovery cannot be achieved with this method [103]. At the module level, PID-s can be prevented by using alternative module components, such as a new type of encapsulant material or glass. For example, using an ionomer instead of EVA as an encapsulant effectively minimizes PID-s because the conductivity of an ionomer is much lower than that of EVA [19, 69]. Selecting module glass with a PID-resistant property, such as absence of sodium or high electrical conductivity, is another simple way to address the



PID-s issue [19, 73, 103]. However, the use of such replacement materials could prove to be expensive and/or could result in lower durability or reliability over the long life of PV modules. We have presented an innovative method to prevent PID-s using surface interruption, which does not require changes in the module components [115]. In this chapter, we present a simple, reliable, and cost-effective method based on our previous work [115] to prevent or minimize PID-s at the module level in the factory or at the system level in the field (patent pending) [116]. This method uses commercially available, thin, and flexible Corning® Willow® Glass sheets or strips on the PV module glass superstrates.

7.2 Experiments

Standard 156 \times 156-mm² p-type monocrystalline silicon solar cells, which are PID-s-susceptible, were obtained from commercial sources and investigated in this study. Each cell was laminated as a one-cell coupon with the same PV module construction materials and structure (glass – EVA – cell – EVA – backsheet) by using a commercial laminator (NPC LM-110x160-s). Typical commercial-grade PV module materials, such as soda-lime solar glass (8 \times 8 inches²), EVA, and TPE backsheet, were chosen for fabricating the one-cell coupons. To simulate an aluminum frame, aluminum tape with conductive adhesive was attached onto the edges of the coupons. The aluminum tape was not attached onto the top edge due to the presence of the positive and negative leads, as shown in Figure 7.1.



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Figure 7.1 Photograph of Two One-Cell Coupons. Red Dotted Areas are the Locations where the Willow Glass (a) Sheet or (b) Strip was placed. (a) coupon A; (b) coupon C.

To replicate and accelerate the PID-s, three environmental conditions were required: temperature, humidity, and voltage. To simplify the test setup, the high humidity that causes high glass surface conductivity can be replaced with aluminum foil on the module glass and aluminum frame. In this case, no humidity-controlled chamber is needed. PID-s stress was applied either at 60 °C and 0% relative humidity (RH) or at 60 °C and 85% RH, with an applied voltage of -600 V on the cell with respect to the aluminum tape at the edges. The 0% RH condition was used when the glass surface was fully covered with aluminum tape, overlapping the aluminum tape at the edges; the 85% RH condition was used when the glass surface was not covered with aluminum tape. All the one-cell coupons were characterized by light current-voltage (I–V), dark I–V, and electroluminescence (EL) imaging before and after the PID stress tests.

Use of flexible Willow Glass has been demonstrated in a variety of flexible electronic applications including displays, touch sensors, lighting, and photovoltaic devices [117]. It is very light, thin ($\sim 100 \ \mu m$), and flexible. Because of its unique alkali-



free borosilicate composition, the glass was considered to be a good candidate for addressing the PID-s issue.



(c)

Figure 7.2 Side Views of Coupon with Willow Glass: (a) Coupon A, (b) Coupon B, (c) Coupon C.

The Willow Glass sheets were attached onto the glass of the test coupons in three different ways. For the first of these, a square Willow Glass sheet (16 cm \times 16 cm) was placed on the one-cell coupon, as shown in Figure 7.2 (a), and then aluminum tape was used to cover the whole surface, including the Willow Glass (coupon A). An existence of



air gap between the Willow Glass and the one-cell coupon glass is possible due to roughness differences of those glass surfaces, and the air gap could increase an electrical resistance causing less front surface conductivity affecting the extent of PID-s testing. To minimize the possible air gap a heavy object was placed on the aluminum tape, and an additional aluminum foil was taped tightening both the object and the one-cell coupon. Because the front aluminum covers the whole cell surface, no humidity was needed to carry out the PID-s test. For the second glass attachment method, the square Willow Glass was placed between the front coupon glass and cell (Coupon B). The module laminator was used to encapsulate the Willow Glass. An additional EVA sheet was utilized for this encapsulation, which has the structure (front glass – EVA – Willow Glass - EVA - cell - EVA - backsheet) shown in Figure 7.2 (b). In the third glass attachment method, aluminum tape was used to attach a rectangular Willow Glass strip (17.5 cm \times 2 cm) around the edges (coupon C). Only half of the glass strips were covered by the edge aluminum tape, as shown in Figure 7.1 (b) and Figure 7.2 (c). Further, various commercially available products were tested as alternative materials for this edge interruption concept. For example, hydrophobic spray (manufacturers A and B) or ionomer were applied to around the edges where the Willow Glass was placed.

Coupons B and C had no aluminum tape on the front surface; thus, 85% RH was used while applying the PID-s stress to these samples. Reference coupons without Willow Glass were also fabricated and the PID-s setup for these was kept identical to the test conditions employed for a given coupon. Thus, 0% RH was used for reference coupon A, and 85% RH used for that of coupons B and C.



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7.3 Square Sheet Willow Glass for Surface Interruption



Figure 7.3 EL Image before/after PID-s: (a) Coupon A Initial, (b) Coupon A after PID-s, (c) Reference Coupon Initial, (d) Reference Coupon after PID-s. PID Stress Applied at 60 °C/0% RH (with Aluminum Tape Front Covered), –600 V, 96 h

As shown in Figure 7.3 (a) and (b), coupon A showed no PID-s in the area where the Willow Glass film/sheet was placed, whereas the reference coupon experienced PID-s resulting in 20% P_{max} decrease all over the cell area. The edge of coupon A was, however, observed to be a bit darker (shunting) after the PID-s stress application when compared with its initial state. The dark shunted area of Coupon A caused a decrease of shunt resistance leading to 5% P_{max} degradation, as shown in Table 7.1. Contact between the front aluminum tape and the gaps between the Willow Glass and edge aluminum tape caused this PID-s progression. Thus, PID-s could be prevented if the gaps were fully insulated from the front aluminum tape in no humidity condition (with front aluminum tape covered). To avoid or minimize the humidity ingress issue between the Willow



Glass and module glass, the Willow Glass could be fixed with an interpenetrating bonding material, for example an ionomer as indicated in Section 7.5.

		$P_{max}(W)$	FF (%)	$R_{shunt}\left(\Omega ight)$
Correct	Initial	3.87	70.4	221
Coupon A	PID 96h	3.68	67.6	9.01
Reference	Initial	3.79	70.4	252
Coupon	PID 96h	3.03	58.3	1.01

 Table 7.1
 Cell Parameter before and after PID-s Stress Testing

Because Willow Glass is thin, light weight, and has acceptable transmittance and exceptionally lower conductivity than common soda-lime glass, it can be applied not only on the front surface of the PV module glass (Coupon A) but also underneath the glass (Coupon B), as shown in Figure 7.2 (b). PID-s stressing was carried out on this one-cell coupon (Coupon B) under 60 °C/85% RH, -600 V conditions for 96 h. In addition, a PID-s-free EVA one-cell coupon (with no Willow Glass but commercially available PID-s-free EVA instead of common EVA) also was PID-s stressed to compare the PID-s-resistance. Whereas coupon B showed extremely strong PID-s-resistance, the PID-s-free EVA coupon ("Coupon B-2") suffered from PID-s that resulted in more than a 10% power decrease, as shown in Figure 7.4. We also performed a PID-s test for a coupon having a commercially available PID-s-free cell with regular EVA, and even this cell exhibited PID-s with a power loss greater than 30% in 96 h. These alleged PID-s-free EVA and cells would both fail the PID draft standard [51]. It is possible that the failure of these PID-s-free EVA and cells might be attributed to non-optimized lamination

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conditions for those components. Nevertheless, it is suggested that PID-s-free EVA/cells available in the open market should be carefully evaluated when replacing the existing components to safely avoid PID-s.



Figure 7.4 PID-s Progress under 60 °C/85% RH, -600 V for 96 h: (a) A Coupon with the Willow Glass, (b) A Coupon with PID-s-Free EVA (no Willow Glass).

Table 7.2Cell Parameter Relative Differences Due to The Additional Willow GlassSheet and EVA Layer

	Isc (%)	V _{oc} (%)	I _{mp} (%)	V _{mp} (%)	P _{max} (%)	FF (%)	Efficiency (%)
Before PID	-1.71	-0.00	-1.46	-0.48	-1.95	-0.14	-1.96
After PID 96h	0.74	0.34	0.17	0.77	0.95	-0.14	0.92



It was expected that adding an additional EVA layer and the Willow Glass between the solar glass and the cell might result in a decrease in the transmittance. The first row in Table 7.2 shows the cell performance loss due to the additional layers in the coupon before PID-s stressing. The data comprising this row was created by measuring I-V performance of nearly identical cells before and after lamination. The decrease of transmittance before PID-s stressing due to the additional layers causes short circuit current (I_{sc}) loss, which results in an efficiency loss (-2% before PID-s). Interestingly, the efficiency (relative) was rather improved from -2% to 1% due to an I_{sc} increase after the 96-h PID-s stress. It is suspected that the test condition (60 °C and 96 h) leads to transmittance improvement of the EVA, which could be caused by better curing of the EVA. Also, a freshness of the EVA could affect the lamination quality as the manufacturer states in the specification sheet. There were limitations in keeping the EVA fresh in ASU Solar Power Laboratory since our laboratory was not designed for commercial PV manufacturing. This might suggest that the lamination process should be optimized to minimize I_{sc} loss or even improve I_{sc} when the Willow Glass is applied below the PV module glass.

The method used in Coupon A and B showed great PID resistance using the Willow Glass. However, Coupon B method has a few disadvantages in applying them on the PV modules operating in a real field. First, overall output power will decrease due to a lower transmittance from additional bonding layer and Willow Glass. Secondly, cost could be considerably higher when the Willow Glass covering all the cells in a PV module is used on the PV module surface or placed between the module glass and an additional EVA. The Coupon B method requiring large surface area of Willow Glass



could lead to marginal or considerable increase in the module manufacturing cost as compared to the other method using alternative encapsulant materials proven to be PIDresistant, such as polyolefin [70, 118-120] or ionomer [68, 69], due to additional process and materials. In order to address these issues, the amount of these additional materials covering the cells/module should be minimized. These issues are addressed by using edge interruption method using Willow Glass strips, which is presented in Sections 7.4 and 7.5. The edge interruption method which is used in Coupon C is much more cost effective than Coupon A or B since the required area of Willow Glass is much less than 10% of the additional material used in Coupon A or B. Moreover, there is no transmittance loss on to the cells since the strips are applied only around the module edges. Therefore, we anticipate that the Coupon C method should have a higher market penetration as compared to the Coupon B method.

An illustrative PID-s circuit with high humidity or rain for the field-installed PV modules can be drawn as shown in Figure 7.5. The surface interruption method demonstrated in this section is maximized by increasing the R_{front} surface as schematically depicted in the circuit diagram shown in Figure 7.5. Willow Glass above the coupon glass provides an extremely high R_{front} surface, which interrupted PID-s. The Willow glass below the coupon glass also increases the resistance between the front glass and the cell so that PID-s is interrupted. In the literature, it has been reported that PID-s can be prevented/reduced by increasing R_{glass} [19, 73] or $R_{encapsulant}$ [32, 69, 70]. An advantage of using the Willow Glass when compared with these PID-s prevention techniques is that there is no need to change potentially PID-s-susceptible PV module materials or cells, which are already used in the module construction. PID-s could be



simply minimized or eliminated even after manufacturing the modules, by simply adding the thin flexible Willow Glass in the PV module structure/construction. In other words, Willow Glass can be used as an insulating or PID-s circuit interrupting barrier to block sodium transport to the cells even if common soda-lime glass is used or PID-s-susceptible cell is used in the module structure.



Figure 7.5 Cross-section of Typical PV Module that shows the Resistance of PV Module Materials Affecting PID-s-Susceptibility.

7.4 Rectangular Strip Willow Glass for Edge Interruption

Using a conceptual setup, previous researchers of this research group reported that PID-s can be prevented or mitigated by interrupting the surface continuity near the frame edge of the PV module [121, 122]. In this method, the circuit shown in Figure 7.5 is conceptually interrupted by a very high value of R_{edge} , thereby preventing PID-s without any transmittance loss to the cells as they are attached away from the cell at the frame



edge. To prove this conceptual approach through the use of a physical material as a circuit interrupter, we applied Willow Glass strips/films to a one-cell coupon to increase the R_{edge}. A portion of the Willow Glass strip was fixed by edge aluminum tape (see Figure 7.2 (c)); thus, the Willow Glass was in direct contact with the coupon glass. The results shown in Figure 7.6 and Figure 7.7 indicate that PID-s can be physically prevented due to near-edge surface disruption caused by the Willow Glass strips. Coupon C showed nearly no degradation in terms of maximum power (Pmax), whereas the Pmax of the reference coupon showed a power loss of 10% at standard test conditions (STC) and 40% at low irradiance, as shown in Figure 7.7 (c). It should be noted that the bottom of the cell in coupon C was slightly affected by PID-s, as shown in the EL image in Figure 7.6 (b). In that area, there was a decrease of about 55% in shunt resistance (R_{sh}), as shown in Figure 7.7 (c). These effects were caused by the unintended conductive path resulting from water ingress between the Willow Glass and the coupon glass during the PID-s stress test. This conductive path can be prevented by avoiding the water ingress with the use of an improved method presented in Section 7.5 for affixing the Willow Glass at the edges of the coupon glass.





Figure 7.6 EL Image before/after PID-s: (a) Coupon C Initial, (b) Coupon C after PID-s, (c) Reference Coupon Initial, (d) Reference Coupon after PID-s.





(c)

Figure 7.7 I–V Results before/after PID: (a) Coupon C, (b) Reference Coupon, (c) Several Important I–V Parameters Regarding PID.



P_{max} (STC)

7.5 Ionomer Bonded Willow Glass Strip for Edge Interruption

Since the edge interruption method in preventing PID-s was proved to be working, it was expected that any high R_{edge} materials that are interrupting the circuit should work as well. One of the candidates studied was a hydrophobic material, which could interrupt the circuit by repelling water near the coupon edges. The hydrophobic material chosen from two different manufacturers was sprayed around the edges of two coupons, and then PID-s stressing was carried out at 60 °C/85% RH, -600 V for 5 h. Both coupons exhibited PID-s. This could be attributable to pinholes or cracks in the hydrophobic layer, as shown in Figure 7.8. The 85% RH in a weathering chamber is maintained by mist, and the mist could possibly permeate into the pinholes/cracks, which facilitate creating a conducting path to the front coupon glass. It is suggested that the hydrophobic materials for the edge interruption method should not have those defects to effectively prevent PID-s if they are applied to module edges. Electrical insulating spray was also tried to increase the R_{edge}. According to the manufacturer of the insulating spray, it is not hydrophobic, but it is used to protect surfaces against weather, moisture, corrosion, oil, alkalies, and acids. The aforementioned PID-s stress conditions were applied to the coupon with the insulating spray, and it was determined that this material also did not work in preventing PID-s. It showed a rather stronger degradation in power than the coupon with the hydrophobic layer. It is suspected that the insulating layer created by spraying also has the pinholes. Another material we found for this edge interruption method was ionomer. The ionomer has a high bulk resistivity and has been known as a good insulation material for PV modules [69]. Ionomer strips the same size as the Willow Glass strips were applied around the one-cell coupon edges using the PV module



laminator. After PID-s stressing at 60 °C/85% RH, -600 V for 5 h, the coupon showed only 1% power decrease while a coupon having no edge interruption showed a power decrease >10%. The ionomer strip was then used as a bonding material in fixing the Willow Glass strip onto the coupon glass. PID-s results of the coupon with Willow Glass bonded by a thin ionomer layer showed only approximately 1% power decrease. As shown in Figure 7.9, there are nearly no changes in I-V curves and no significant shunting spots based on EL images. It is expected that the Willow Glass strip with ionomer would provide reliable PID-s prevention since the Willow Glass provides a hermetic good barrier for the well-demonstrated Willow Glass circuit interrupter. In addition, the Willow Glass protects the ionomer from deterioration caused by direct atmosphere exposure, such as UV, humidity and soiling. However, application of Coupon C method in the field would be challenging as it currently requires a lamination process of Willow Glass strips with underlying ionomer layer. It is, therefore, anticipated that the application of Coupon C method in the field without lamination process requires additional research to successfully implement this method in the field. Nevertheless, this work still demonstrates that the ionomer bonded Willow Glass strips near the frame edges on the glass surface can prevent the PID-s in the long term field conditions through the removal of conductive path resulting from water ingress between Willow Glass and module glass.





Figure 7.8 Optical Microscope Image (10x) Showing Cracks of the Hydrophobic Layer Sprayed around the One-Cell Coupon Edges.



(a)



Figure 7.9 PID-s Test Results of the Coupon with Willow Glass + Ionomer: (a) I–V Results before/after PID-s, (b) EL Image before PID-s, (c) EL Image after PID-s.



7.6 Summary

In this chapter, we present a simple method that uses thin flexible/film Corning® Willow® Glass on the surface of the glass superstrate to prevent PID-s. Adding a Willow Glass layer onto the glass superstrate effectively minimized or eliminated PID-s. One of innovative advantages is that Willow Glass could be applied onto the glass superstrates of modules not only during manufacturing but also in the field by taking advantage of edge interruption. Currently manufactured PV modules are expected to mostly contain the PID-s-free cells but a large fraction of the p-base crystalline silicon PV modules installed over the last 10 years might have a potential susceptibility to PID-s. Applying this edge interruption method to those field-installed modules would be a simple and low-cost way to prevent or minimize PID-s progression. The surface disruption method presented in this work was demonstrated to be effective. Further experiments are in progress to apply this method to full-size commercial PV modules.



Chapter 8

CONCLUSION AND FUTURE WORK

8.1 Conclusion

In this dissertation, various characteristics of PID, especially recovery characteristics, were studied. No complete recovery in terms of power at low irradiance, shunt resistance, and QE at short wavelength was achieved regardless of recovery methods, such as applying high temperature or reverse potential, although power at standard test condition (high irradiance) showed near full recovery. Low recovery of shunt resistance in PID-affected solar cells could cause detrimental effects to the modules if the cells in a module were to operate under shaded/reverse-bias condition with failed bypass diodes. Therefore, this work recommends that the type of PID recovery method should be carefully investigated as a long-term solution to modules exhibiting PID. QE loss, which could be one of the PID effects in the short wavelength range, was modelled by increasing the front surface recombination velocity and verified by QE measurements on bare cells, and this was carried out utilizing a new PID testing method using cells that had not gone through a lamination process.

At the cell level, PID could be avoided by the use of high phosphorous concentration emitters (low sheet resistance: ~ 60 Ω /sq) even if the SiN_x ARC is PID-susceptible in p-type crystalline silicon solar cells. Also, PID-susceptible cells can be screened by using ILIT with simplified sample structure during the cell fabrication process. At the module level, PID could be prevented by a leakage current circuit disruption method. The circuit was effectively interrupted by using either Corning Willow Glass sheets or Corning Willow Glass strips. The Willow Glass sheets could be



applied below the module glass or above the module glass during the module manufacturing process. A PV module having a Willow Glass strip attached peripherally around the module edges showed no PID as well as no transmittance loss while the one incorporating a blanket Willow Glass sheet has exhibited transmittance loss. This method can be applied to eliminate the PID issue in the PID-susceptible crystalline silicon PV modules already installed in a field. Therefore, there is no need to replace those modules from the field to avoid PID in future.

8.2 Future Work

The effect of the surface phosphorous concentration in the emitter on PID has been presented in this dissertation. Cells with high surface phosphorous concentration emitters clearly prevented or delayed the PID progress while ones with low surface phosphorous concentration emitters experienced PID. However, it is unclear how it physically works. For the emitter analysis, PID-susceptible (low surface phosphorous concentration emitter) cells fabricated from semiconductor polished wafers for SIMS and ECV measurements were PID stressed; however, no clear PID was observed. Therefore, surface analysis for cells fabricated by textured solar wafers are needed, but it is challenging to measure SIMS or ECV on textured solar cells. The PID prevention method using the Willow Glass strip has been demonstrated for one-cell laminated coupons. It is recommended to apply the PID prevention based upon Willow Glass strips to commercial-size PV modules.



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